ELECTRONS IN ATOMS

A. Matching

Allotels words before in Columns II with the correct alexer(polen in Collamn A, White the letter of the correct iteration in the line.

		Column A		Column B	
-	1	the ways in which electrons are arranged or naclei of strong	word the A	Author principle	
	2.	the ejection of electrons by metals when lig- on them	br.shines B.	CAMPIAN	
	36	the region around the saudous of an atom velection is likely to be morting.	there as6	photosilectric effect	
	100	An atomic exhibit map describe at most two	electrimis. d.	attenute orbitals	
	*	the seglons within which electrons have the probability of heling found	bighesi e.	Planti evaluation principle	
_	4.	When electrons eccupy orbitals of equal en- electron entits such orbital antil sili the orb- rosistic one electron.		manage bread	
-	70	the amount of mangy required to more on a from its passess energy level to the next hig		Humb's male	
	100	Right quarte.	16.	electron configurations	
	*	Electronic notes substate of inevest energy file	. L	quantum mechanical model	
	1000	the madero description of the electrons in	energy 3.	photone	
B. Mul	letic	le Choice			
Charges at	ne de	at exposer could serine list letter we che libet.			
	116	 iii. Bolar's contribution to the development of stancic structum. a. was referred to as the "plum problets model." b. was the discovery that electrons summed a deter nucleus. c. was proposed that electrons travel to circular-orbits around the trackets. d. is the quantum mechanical model. 			
	12.	What is the total number of orbitals in the talls. I so 1 so 4.		engy level?	

<u>Pearson Education Chapter 5 Electrons In Atoms Answer</u> <u>Key</u>

Hussin A.Rothana

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