# SUMMER HOLIDAY WORKSHEET

# CHAPTER 01: - SOME BASIC CONCEPTS OF CHEMISTRY

- Out of 1m NaOH solution and 1M NaOH solution, which one is concentrated?
- 2. How many atoms and molecules are present in 48g of ozone?
- 1L of a solution of oxalic acid [(COOH)<sub>2</sub>. 2H<sub>2</sub>O] contains 2.52g of acid.
   Calculate the molarity of the solution.
- 4. An organic substance containing C, H and O gave the following percentage composition as C =40.687%, H =5.085% and O=54.228%. Calculate the empirical and molecular formula of the salt. Given molecular mass = 118g/mol.
- 5. Vitamin C is known to contain  $1.29x10^{24}$  hydrogen atoms. Calculate the number of moles of hydrogen atoms.
- Determine the empirical formula of an oxide of iron which has 69.9% iron and 30.1% oxygen by mass. (Atomic mass of Fe=55.85u, O=16u).
- 7. Conc.  $\rm H_2SO_4$  is 98% by mass and has density 1.84 g/cm<sup>3</sup>. What volume of concentrated acid is required to make 5.0 L of 0.5 M  $\rm H_2SO_4$  solution?
- 8. What mass of slaked lime would be required to decompose completely 4 grams of ammonium chloride and what would be the mass of each product? Ca (OH)<sub>2</sub> + 2NH<sub>4</sub>Cℓ → CaCℓ<sub>2</sub> + 2NH<sub>3</sub> + 2H<sub>2</sub>O
- The density of 1M solution of NaNO<sub>3</sub> has density 1.25 g/cm<sup>3</sup>.
   Calculate the molality. (Molar mass of NaNO<sub>3</sub> = 85g/mol)
- 10.80g of hydrogen and 80g of oxygen react with each other to produce water according to the equation:  $H_2(g)+O_2(g)\to 2H_2O(g)$ 
  - (i) Identify the limiting reagent.
  - (ii) Calculate the amount of water formed.
  - (iii) Find the amount of the other reactant that remains unreacted.
- 11.An organic compound containing carbon hydrogen and oxygen gave the following composition. Carbon 54.55%, hydrogen 9.09% and the

# **Pearson Education Chemisrty Ch 11 1 Key**

**B** Lingard

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Organic Chemistry Pierre Vogel, Kendall N. Houk, 2019-07-30 Provides the background tools and models required to understand organic synthesis and plan chemical reactions more efficiently Knowledge of physical chemistry is essential for achieving successful chemical reactions in organic chemistry Chemists must be competent in a range of areas to understand organic synthesis Organic Chemistry provides the methods models and tools necessary to fully comprehend organic reactions Written by two internationally recognized experts in the field this much needed textbook fills a gap in current literature on physical organic chemistry Rigorous yet straightforward chapters first examine chemical equilibria thermodynamics reaction rates and mechanisms and molecular orbital theory providing readers with a strong foundation in physical organic chemistry Subsequent chapters demonstrate various reactions involving organic organometallic and biochemical reactants and catalysts Throughout the text numerous questions and exercises over 800 in total help readers strengthen their comprehension of the subject and highlight key points of learning The companion Organic Chemistry Workbook contains complete references and answers to every question in this text A much needed resource for students and working chemists alike this text Presents models that establish if a reaction is possible estimate how long it will take and determine its properties Describes reactions with broad practical value in synthesis and biology such as C C coupling reactions pericyclic reactions and catalytic reactions Enables readers to plan chemical reactions more efficiently Features clear illustrations figures and tables With a Foreword by Nobel Prize Laureate Robert H Grubbs Organic Chemistry Theory Reactivity and Mechanisms in Modern Synthesis is an ideal textbook for students and instructors of chemistry and a valuable work of reference for organic chemists physical chemists and chemical engineers **Chemistry: The Key to our Sustainable** Future Minu Gupta Bhowon, Sabina Jhaumeer-Laulloo, Henri Li Kam Wah, Ponnadurai Ramasami, 2013-11-08 Chemistry The Key to our Sustainable Future is a collection of selected contributed papers by participants of the International Conference on Pure and Applied Chemistry ICPAC 2012 on the theme of Chemistry The Key for our Future held in Mauritius in July 2012 In light of the significant contribution of chemistry to benefit of mankind this book is a collection of recent results generated from research in chemistry and interdisciplinary areas It covers topics ranging from nanotechnology natural product chemistry to analytical and environmental chemistry Chemistry The Key to our Sustainable Future is written for graduates postgraduates researchers in industry and academia who have an interest in the fields ranging from fundamental to applied **Measurement and Probability** Giovanni Battista Rossi, 2014-05-19 Measurement plays a fundamental role chemistry both in physical and behavioral sciences as well as in engineering and technology it is the link between abstract models and empirical reality and is a privileged method of gathering information from the real world Is it possible to develop a single theory of measurement for the various domains of science and technology in which measurement is involved This book takes the challenge by addressing the following main issues What is the meaning of measurement How do we measure What can be measured A theoretical framework that could truly be shared by scientists in different fields ranging from physics and engineering to psychology is developed The future in fact will require greater collaboration between science and technology and between different sciences Measurement which played a key role in the birth of modern science can act as an essential interdisciplinary tool and language for this new scenario A sound theoretical basis for addressing key problems in measurement is provided These include perceptual measurement the evaluation of uncertainty the evaluation of inter comparisons the analysis of risks in decision making and the characterization of dynamical measurement Currently increasing attention is paid to these issues due to their scientific technical economic and social impact The book proposes a unified probabilistic approach to them which may allow more rational and effective solutions to be reached Great care was taken to make the text as accessible as possible in several ways Firstly by giving preference to as interdisciplinary a terminology as possible secondly by carefully defining and discussing all key terms This ensures that a wide readership including people from different mathematical backgrounds and different understandings of measurement can all benefit from this work Concerning mathematics all the main results are preceded by intuitive discussions and illustrated by simple examples Moreover precise proofs are always included in order to enable the more demanding readers to make conscious and creative use of these ideas and also to develop new ones The book demonstrates that measurement which is commonly understood to be a merely experimental matter poses theoretical questions which are no less challenging than those arising in other apparently more theoretical disciplines Knowledge Visualization and Visual Literacy in Science Education Ursyn, Anna, 2016-05-31 Effective communication within learning environments is a pivotal aspect to students success By enhancing abstract concepts with visual media students can achieve a higher level of retention and better understand the presented information Knowledge Visualization and Visual Literacy in Science Education is an authoritative reference source for the latest scholarly research on the implementation of visual images aids and graphics in classroom settings and focuses on how these methods stimulate critical thinking in students Highlighting concepts relating to cognition communication and computing this book is 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G. R. Briggs, 2009 Engineering Education John Heywood, 2006-01-24 A synthesis of nearly 2 000 articles to help make engineers better educators While a significant body of knowledge has evolved in the field of engineering education over the years much of the published information has been restricted to scholarly journals and has not found a broad audience This publication rectifies that situation by reviewing the findings of nearly 2 000 scholarly articles to help engineers become better educators devise more effective curricula and be more effective leaders and advocates in curriculum and research development. The author's first objective is to provide an illustrative review of research and development in engineering education since 1960 His second objective is with the examples given to encourage the practice of classroom assessment and research and his third objective is to promote the idea of curriculum leadership The publication is divided into four main

parts Part I demonstrates how the underpinnings of education history philosophy psychology sociology determine the aims and objectives of the curriculum and the curriculum's internal structure which integrates assessment content teaching and learning Part II focuses on the curriculum itself considering such key issues as content organization trends and change A chapter on interdisciplinary and integrated study and a chapter on project and problem based models of curriculum are included Part III examines problem solving creativity and design Part IV delves into teaching assessment and evaluation beginning with a chapter on the lecture cooperative learning and teamwork The book ends with a brief insightful forecast of the future of engineering education Because this is a practical tool and reference for engineers each chapter is self contained and may be read independently of the others Unlike other works in engineering education which are generally intended for educational researchers this publication is written not only for researchers in the field of engineering education but also for all engineers who teach All readers acquire a host of practical skills and knowledge in the fields of learning philosophy sociology and history as they specifically apply to the process of engineering curriculum improvement and evaluation

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In a digital era where connections and knowledge reign supreme, the enchanting power of language has be apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Pearson Education Chemisrty Ch 11 1 Key**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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