# Chemistry Unit 8 Worksheet 4 Samples of Every Kind of Problem

On a separate sheet of paper, write a complete solution to each of the problems below. Follow the procedure outlined in class. Be sure to circle your final answer.

 Calculate the number of moles of potassium chlorate, KClO<sub>3</sub> (s), that must decompose to produce potassium chloride, KCl (s), and 1.8 moles of oxygen gas.

Equation:	2 KClO <sub>3</sub>	→ 2 KCI +	3 02
Before:	1.2	•	0
Change:	-1.2	+1.2	+1.8
After:	0	1.2	1.8

1.2 moles KClO<sub>2</sub> needed

2. In a single displacement reaction, magnesium metal reacts with hydrochloric acid to produce magnesium chloride and hydrogen gas. How many moles of hydrochloric acid are needed to completely react with 2.43 g of magnesium?

Equations	Nig +	2 HCl →	MgCl <sub>1</sub> +	HI a
Before:	0.100	7	0	0
Change:	-0.100	-0.200 +	0.100 +	0.100
Afters		7	0.100	0.100

0.200 moles HCI needed

3. Ethane, C<sub>2</sub>H<sub>6</sub> reacts with oxygen gas to produce carbon dioxide gas and water vapor. What mass of oxygen gas is required to react with 2.20 moles of ethane?

Equation:	2 C2H6	+ 7 O <sub>2</sub>	→ 4 CO;	+ 6.H <sub>2</sub> O
Before:	2.20	76.76	0	
Changes	-2.20	- 7.70	+ 4,40	+ 6.60
After:	0	3.5	4.40	6.60

O<sub>2</sub> needed: 7.70 mole x 32.0 g = 246 g O<sub>2</sub> required 1 mole

 Determine the mass of sodium nitrate produced when 0.73 g of nickel (II) nitrate reacts with sodium hydroxide according to the following equation:

Ni(NO<sub>3</sub>)<sub>2</sub> + 2 NaOH --> Ni(OH)<sub>2</sub> + 2 NaNO<sub>3</sub>

0.73 g x 1 mole = .0040 mole 183 g

Equation:	Ni(NO <sub>3</sub> ):	2 NaOH	→ Ni(OH)2	+ 2 NaNO3
Before:	0.0040	XS	0	0
Change:	- 0.0040	- 0.0080	+0.0040	+0.0080
After:	0	NS	0.0040	0.0080

NaNO, produced: 0.0080 mole x 85.0 g = 0.68 g I mole

# **Pearson Education Chemistry Worksheet**

Bridget Mulvey, Adam Scott Ward, Anne J. Jefferson, Jerad Bales

#### **Pearson Education Chemistry Worksheet:**

Learning with Understanding in the Chemistry Classroom Iztok Devetak, Saša Aleksij Glažar, 2014-01-14 This volume offers a critical examination of a variety of conceptual approaches to teaching and learning chemistry in the school classroom Presenting up to date research and theory and featuring contributions by respected academics on several continents it explores ways of making knowledge meaningful and relevant to students as well as strategies for effectively communicating the core concepts essential for developing a robust understanding of the subject Structured in three sections the contents deal first with teaching and learning chemistry discussing general issues and pedagogical strategies using macro sub micro and symbolic representations of chemical concepts Researchers also describe new and productive teaching strategies The second section examines specific approaches that foster learning with understanding focusing on techniques such as cooperative learning presentations laboratory activities multimedia simulations and role playing in forensic chemistry classes The final part of the book details learner centered active chemistry learning methods active computer aided learning and trainee chemistry teachers use of student centered learning during their pre service education Comprehensive and highly relevant this new publication makes a significant contribution to the continuing task of making chemistry classes engaging and effective Research in Chemistry Education Liliana Mammino, Jan Apotheker, 2021-05-17 This volume emphasizes the role of chemical education for development and in particular for sustainable development in Africa by sharing experiences among specialists across the African continent and with specialists from other continents It considers all areas and levels of chemistry education gives specific attention to known major challenges and encourages explorations of novel approaches The chapters in this book describe new teaching approaches approach explorations and in class activities analyse educational challenges and possible ways of addressing them and explore cross discipline possibilities and their potential benefits for chemistry education This makes the volume an up to date compendium for chemistry educators and educational researchers <u>Teaching Science Online</u> Dietmar Kennepohl, 2023-07-03 With the increasing focus on science education worldwide growing attention is being paid to how science is taught Educators in science and science related disciplines are recognizing that distance delivery opens up new opportunities for delivering information providing interactivity collaborative opportunities and feedback as well as for increasing access for students This book presents the guidance of expert science educators from the US and from around the globe They describe key concepts delivery modes and emerging technologies and offer models of practice The book places particular emphasis on experimentation lab and field work as they are fundamentally part of the education in most scientific disciplines Chapters include Discipline methodology and teaching strategies in the specific areas of physics biology chemistry and earth sciences An overview of the important and appropriate learning technologies ICTs for each major science Best practices for establishing and maintaining a successful course online Insights and tips for handling practical components like laboratories and field work Coverage of breaking topics including

MOOCs learning analytics open educational resources and m learning Strategies for engaging your students online The School Science Review ,2002 Quantification, Validation and Uncertainty in Analytical Sciences Max Feinberg, Serge Rudaz, 2024-02-16 Quantification Validation and Uncertainty in Analytical Sciences Companion guide explaining all processes in measuring uncertainty in quantitative analytical results Quantification Validation and Uncertainty in Analytical Sciences provides basic and expert knowledge by building on the sequence of operations starting from the quantification in analytical sciences by defining the analyte and linking it to the calibration function Proposing a comprehensive approach to MU Measurement Uncertainty estimation it empowers the reader to apply Method Accuracy Profile MAP efficiently as a statistical tool in measuring uncertainty The text elucidates several examples and template worksheets explaining the theoretical aspects of the procedure and includes novel method validation procedures that can accurately estimate the data obtained in measurements It also enables the reader to provide practical insights to improve decision making by accurately evaluating and comparing different analytical methods Brings together an interdisciplinary approach with statistical tools and algorithms applied in analytical chemistry and written by two international experts with long standing experience in the field of Analytical measurements and Uncertainty Quantification Validation and Uncertainty in Analytical Sciences includes information on The know how of methods in an analytical laboratory effective usage of a spurious measurement and methods to estimate errors Quantification calibration precision trueness MAP addons estimating MU for analytical sciences and uncertainty functions Employing measurement uncertainty sampling uncertainty quantification limits and sample conformity assessment Decision making uncertainty and standard addition method and accuracy profile for method comparison Quantification Validation and Uncertainty in Analytical Sciences is an ideal resource for every individual quantifying or studying analytes With several chapters dedicated to MU s practical use in decision making demonstrating its advantages the book is primarily intended for professional analysts although researchers and students will also find it of interest

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\*\*Digging Deeper\*\* M. Gail Hickey, Jeremiah Clabough, 2017-06-01 Experts in social studies\*\*

education and gifted education share teacher tested strategies for differentiating social studies in K 12 classrooms Chapter authors showcase best practice and research based lessons and activities that enrich and expand social studies instruction while building K 12 students critical and creative thinking Each chapter contains two or more teacher tested lessons or activities linking social studies content and concepts to the standards and recommendations of the National Association for Gifted Children NAGC and National Council for the Social Studies NCSS This edited volume is targeted toward K 12 teachers and administrators gifted education coordinators and consultants parents of gifted children social studies methods instructors and central office administrators Each chapter contains activities that can be adapted and replicated in teachers classrooms Chapters focus on significant social studies topics such as civic education historical thinking drama and teaching with primary sources Each topic is approached in ways that meet the needs of gifted education students Through its emphasis on critical thinking inquiry based instruction and higher order thinking skills activities and lessons in the book challenge K 12 educators to raise the bar for classroom instruction in ways that improve opportunities of learning for all Concept-Based Curriculum and Instruction for the Thinking Classroom H. Lynn Erickson, Lois A. students Lanning, Rachel French, 2017-02-02 Develop students critical thinking abstract reasoning and creative learning skills with concept based teaching Take learning beyond the facts with a teaching approach that develops conceptual thinking and problem solving skills A Concept Based curriculum recaptures students innate curiosity about the world and provides the thrilling feeling of using one s mind well Concept Based teachers will learn how to Meet the demands of rigorous academic standards Use the Structure of Knowledge and Process when designing disciplinary units Engage students in inquiry through inductive teaching Identify conceptual lenses and craft quality generalizations The Software Encyclopedia ,1986

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