

Chemical Reactions



REACTIONS

11.1 Describing Chemical Reactions



Essential Understanding Chemical reactions are represented by balanced chemical equations.

Lesson Summary

Introduction to Chemical Equations. A chemical equation uses symbols, and sometimes words, to show the reactants and products of a chemical reaction.

- A skeleton equation uses chemical formulas to represent reactants and products, but it does not indicate the relative amounts of each.
- Anything that enters into a reaction, such as heat or a catalyst, but is not a reactant or product is shown above or below the yields arrow in the equation.

Balancing Chemical Equations Chemical equations are balanced to show that mass is conserved during chemical reactions.

- A balanced equation shows the relative amounts of reactants and products, and it contains equal numbers of each type of atom on both sides of the equation.
- Chemical equations are balanced by using coefficients in front of the chemical formulas for the reactants and the products in a skeleton equation.



BUILD Math Skills

Boloncing Equations All chemical equations must be balanced because of the law of conservation of mass, which states that matter cannot be created or destroyed. So, the number of atoms that you start with at the beginning of the reaction must equal the number of atoms that you end up with.

For example, the reaction $2\text{Mg} + O_s \rightarrow 2\text{MgO}$ follows the law of conservation of mass because you start with 2 magnesium atoms and 2 oxygen atoms and you end up with 2 magnesium atoms and 2 oxygen atoms. You can think of the \rightarrow as an = sign.



Turn the page to learn more about balancing equations.

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H. Scott Fogler

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Focus on Physical Science California Edition Michael J. Padilla,2007 Furfural: An Entry Point Of Lignocellulose In Biorefineries To Produce Renewable Chemicals, Polymers, And Biofuels Manuel Lopez Granados, David Martin Alonso, 2018-06-19 There is a wide consensus that furfural a renewable commodity currently obtained from lignocellulosic agro residues with a production volume of around 300 kTon per year is a key feedstock for leveraging lignocellulosic residues in future biorefineries Several chemicals are already being manufactured from furfural due to its advantageous production cost Furthermore a vast number of others are also technically viable to produce from oil This book compiles the vast existing information into relevant stages of transformations of furfural as renewable chemicals biofuels and bioresins focusing on the relevant chemical and engineering aspects of processes to obtain them including reactors and catalysis It offers essential information for improving the economic and environmental viability of current commercial applications and upcoming future applications It should be of particular interests to graduate and advanced undergraduate students as well as engineers and academic researchers alike who are working in the field Prentice Hall Science Explorer Michael J. Padilla, 2002

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Taylor, 2019-06-07 This conference proceedings focuses on enabling science and mathematics practitioners and citizens to
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web activity 1 1 2 simple machines practice procedure answer the following questions regarding simple machine systems each question requires proper illustration and annotation including labeling of forces distances direction and unknown values illustrations should consist of basic simple machine functional sketches rather than