

5.4

THE PERIODIC TABLE: ORGANIZING THE ELEMENTS

SECTION REVIEW

Objectives

- Describe the layout of the periodic table
- Identify the position of groups, periods, and the transition metals in the periodic table

Key Terms

- | | | |
|---------------------------|-------------------------|---------------|
| • periodic table | • metals | • non-metals |
| • periods | • alkali metals | • halogens |
| • periodic law | • alkaline earth metals | • noble gases |
| • group | • transition metals | • metalloids |
| • representative elements | • lanthanide elements | |

Part A Completion

Use this newspaper analogy to check your understanding of the concepts and terms discussed in this section. Each blank uses the compound with a name, when possible, or a symbol.

The periodic table organizes the elements into vertical _____ 1. _____ and horizontal _____ 2. _____ in order of increasing _____ 3. _____. The table is constructed so that elements that have similar chemical properties are in the same _____ 4. _____. The elements in Groups 1A through 7A are called the _____ 5. _____. The _____ 6. _____ make up Group 8. The elements in Groups 9A and 10A are interrupted in periods 4 and 5 by the _____ 7. _____ and in the periods 6 and 7 by the _____ 8. _____. The Group 1A elements are called the _____ 9. _____ and the Group 2A elements are called the _____ 10. _____. The non-metals of _____ 11. _____ are called the halogens. Elements with properties that are intermediate between those of metals and non-metals are called _____ 12. _____.

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**Richard G. Lipsey, Christopher T. S.
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