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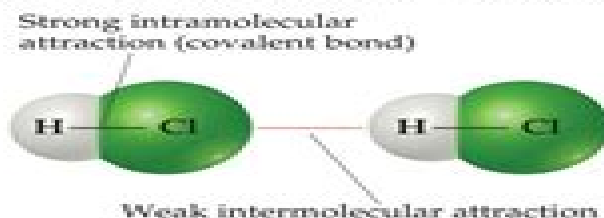
Liquids and Intermolecular Forces

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Hamden, CT

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Intermolecular Forces



- The attractions *between* molecules are not nearly as strong as the **intramolecular** attractions (bonds) that hold compounds together.
- Many physical properties reflect **intermolecular** forces, like boiling points, melting points, viscosity, surface tension, and capillary action.

Intermolecular
Forces

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