Lesson Quiz

Observing Chemical Change

Write the letter of the correct answer on the line at the left.

- Which of the following is true about chemical reactions?
 - A They are accompanied by changes in energy.
 - B They form new substances with new properties.
 - C both A and B
 - D neither A nor B
- Which of the following is NOT a physical property?
 - A melting point
 - B state of matter
 - C density
 - D flammability

- In an endothermic reaction, energy is
 - A absorbed
 - B released
 - C converted to mass
 - D synthesized
- Substances formed as a result of a chemical reaction are called
 - A catalysts
 - B precipitates
 - C products
 - D reactants

If the statement is true, write true. If the statement is false, change the underlined word or words to make the statement true.

- In an exothermic reaction, products have more energy than reactants.
- Water boils at 100°C. This is an example of a chemical property.
- Substances that enter into a chemical reaction are called products.
- The ability to react with oxygen is an example of a <u>chemical</u> property.
- Another name for a chemical change is a chemical <u>bond</u>.
- In a physical change, some of the physical properties of the substance may be altered and the chemical composition remains the same.

Pearson Education Workbook Answers Observing Chemical

Rex M. Heyworth

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Young Scientist Series ICSE Chemistry Work Book 6, Young Scientist Series ICSE Chemistry Work Book 7, Focus on Physical Science California Edition Michael J.

The Software Encyclopedia ,1988 Laboratory Manual of Pharmaceutical Organic Chemistry II Mr. Padilla,2007 Sushant Sudhir Pande, Dr. Prashant D. Aragade, Dr. Niranjan S. Mahajan, Dr. Rahul P. Umbarkar, 2024-02-16 We are very pleased to put forth the Laboratory Manual of Pharmaceutical Organic Chemistry II This manual is prepared as per PCI B Pharm course regulations 2014 and is divided into three sections for laboratory techniques determination of oil values and preparations of organic compounds The methods of all the experiments are added from the recent research papers so that the advancement in the methods or apparatus can be addressed Young Scientist Series ICSE Chemistry 8 Lorraine Pm Science P5/6 Guided Wb Energy, Scott Foresman Science ,2008 Mirza, Krishnakali Gupta, Scientist Series ICSE Chemistry 7 Lorraine Mirza, Krishnakali Gupta, Sif: Chemistry S5n Practical Wb J. G. R. Representations of Nature of Science in School Science Textbooks Christine McDonald, Fouad Briggs, 2002 Abd-El-Khalick, 2017-04-21 Bringing together international research on nature of science NOS representations in science textbooks the unique analyses presented in this volume provides a global perspective on NOS from elementary to college level and discusses the practical implications in various regions across the globe Contributing authors highlight the similarities and differences in NOS representations and provide recommendations for future science textbooks This comprehensive analysis is a definitive reference work for the field of science education **Chemistry 'O' Level** Rex M. Heyworth, 2007 Chemistry insights 'O' level Rex M. Heyworth, 2007 Young Scientist Series ICSE Chemistry 6,

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