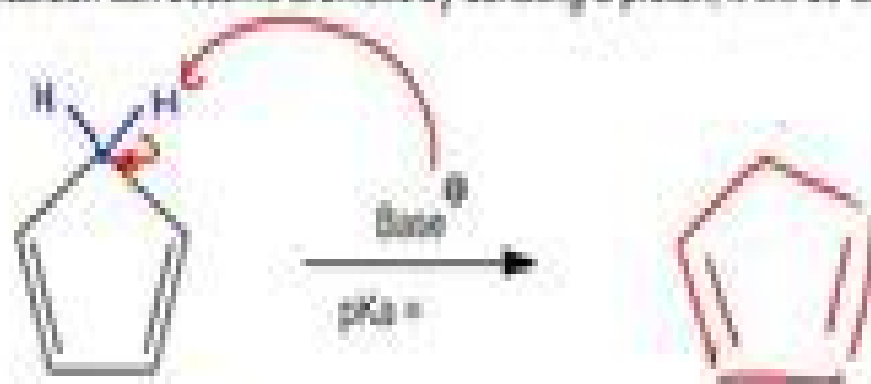


CONCEPT: ACIDITY OF AROMATIC HYDROCARBONS

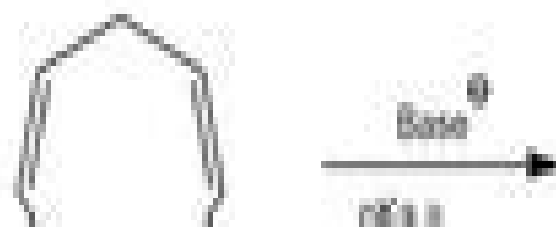
Aromatic hydrocarbons are not naturally acidic. In fact, the pK_a of benzene is 44

- If a hydrocarbon can become aromatic by donating a proton, it will be uniquely acidic. i.e. cyclopentadiene



Non-aromatic

- If a hydrocarbon becomes antiaromatic by donating a proton, it will be uniquely non-acidic. i.e. cycloheptatriene



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