

CHEM 102 – Recitation Week 8 – Answer Key

Ch. 16, 77. A 0.148 M solution of a monoprotic acid has a percent ionization of 1.55%. Determine the acid ionization constant (K_a) for the solution.

Concepts needed: definition of Brønsted-Lowry acids;

definition of monoprotic acids;

definition of percent ionization = how much of the initial amount of acid is ionized, expressed as a percentage (that is, related to 100 units of initial acid);

definition of acid ionization constant, K_a = the equilibrium constant of the ionization of an acid in aqueous solution;

definition of molarity

Answer:

A monoprotic acid will generate one mole of protons per mole of acid in aqueous solution.

The problem says that only 1.55% of the initial 0.148 M acid will be found as ionized in solution:



Since the ionization stoichiometry is 1 to 1,

$$[\text{H}_3\text{O}^+] = 1.55\% \times 0.148 \text{ M} = 2.294 \times 10^{-3} \text{ M} = [\text{A}^-]$$

Calculate the amount of acid that did not ionize:

$$0.148 \text{ M} - 2.294 \times 10^{-3} \text{ M} = 0.145706 \text{ M} = 0.146 \text{ M}$$

If you want to illustrate the ionization process as the ionization equilibrium is reached, you can write an ICE table as follows:

	$\text{HA}_{(aq)}$	$\text{H}_2\text{O}_{(l)}$	\rightleftharpoons	$\text{H}_3\text{O}^+_{(aq)}$	$+$	$\text{A}^-_{(aq)}$
I	0.148	—		0		0
C	-2.294×10^{-3}			$+2.294 \times 10^{-3}$		$+2.294 \times 10^{-3}$
E	0.145706			2.294×10^{-3}		2.294×10^{-3}

$$K_a = [\text{H}_3\text{O}^+][\text{A}^-] / [\text{HA}]_{\text{un-ionized}}$$

$$K_a = \frac{(2.294 \times 10^{-3})^2}{0.145706}$$

$$K_a = 3.61 \times 10^{-5}$$

Pearson Chemistry Answer Key Ch 16 Solutions

Ingmar Persson, Toshio Yamaguchi



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Prentice Hall Chemistry Antony C. Wilbraham, 2008 *Chemistry: The Central Science* Theodore L. Brown, H. Eugene LeMay Jr., Bruce E. Bursten, Catherine Murphy, Patrick Woodward, Steven Langford, Dalius Sagatys, Adrian George, 2013-10-04

If you think you know the Brown LeMay Bursten Chemistry text think again In response to market request we have created the third Australian edition of the US bestseller Chemistry The Central Science An extensive revision has taken this text to new heights Triple checked for scientific accuracy and consistency this edition is a more seamless and cohesive product yet retains the clarity innovative pedagogy functional problem solving and visuals of the previous version All artwork and images are now consistent in quality across the entire text And with a more traditional and logical organisation of the Organic Chemistry content this comprehensive text is the source of all the information and practice problems students are likely to need for conceptual understanding development of problem solving skills reference and test preparation *Chemistry* McMurry, Pearson, 2003-07 **Longman A-level Course in Chemistry** J. G. R. Briggs, 2002-12 **Chemistry O Level Mauritius Edition** Pearson Education, Limited, 2007-06-26 Metal Complexes in Aqueous Solutions Arthur E. Martell, Robert D. Hancock, 2013-06-29

Stability constants are fundamental to understanding the behavior of metal ions in aqueous solution Such understanding is important in a wide variety of areas such as metal ions in biology biomedical applications metal ions in the environment extraction metallurgy food chemistry and metal ions in many industrial processes In spite of this importance it appears that many inorganic chemists have lost an appreciation for the importance of stability constants and the thermodynamic aspects of complex formation with attention focused over the last thirty years on newer areas such as organometallic chemistry This book is an attempt to show the richness of chemistry that can be revealed by stability constants when measured as part of an overall strategy aimed at understanding the complexing properties of a particular ligand or metal ion Thus for example there are numerous crystal structures of the Li ion with crown ethers What do these indicate to us about the chemistry of Li with crown ethers In fact most of these crystal structures are in a sense misleading in that the Li ion forms no complexes or at best very weak complexes with familiar crown ethers such as 12 crown 4 in any known solvent Thus without the stability constants our understanding of the chemistry of a metal ion with any particular ligand must be regarded as incomplete In this book we attempt to show how stability constants can reveal factors in ligand design which could not readily be deduced from any other physical technique Hydrothermal and Supercritical Water Processes Gerd Brunner, 2014-04-04

Hydrothermal and Supercritical Water Processes presents an overview on the properties and applications of water at elevated temperatures and pressures It combines fundamentals with production process aspects Water is an extraordinary substance At elevated temperatures and pressures its properties change dramatically due to the modifications of the molecular structure of bulk water that varies from a stable three dimensional network formed by hydrogen bonds at low and moderate temperatures to an assembly of separated polar water molecules at

high and supercritical temperatures With varying pressure and temperature water is turned from a solvent for ionic species to a solvent for polar and non polar substances This variability and an enhanced reactivity of water have led to many practical applications and to even more research activities related to such areas as energy transfer extraction of functional molecules unique chemical reactions biomass conversion and fuel materials processing destruction of dangerous compounds and recycling of useful ones growth of monolithic crystals and preparation of metallic nanoparticles This book provides an introduction into the wide range of activities that are possible in aqueous mixtures It is organized to facilitate understanding of the main features outlines the main applications and gives access to further information Summarizes fundamental properties of water for engineering applications Compares process and reactor designs Evaluates processes from thermodynamic economic and social impact viewpoints

Metal Ions and Complexes in Solution Ingmar Persson, Toshio Yamaguchi, 2023-12-04 Based on a translated Japanese title published in 2012 this book provides fundamental aspects of experimental and computational methods the properties and structure of solvents ion solvation and equilibria and reactions of metal complexes in solution It includes state of the art details on metal complexes in newly developing sustainable liquids and applications in real life Appealing to researchers working in coordination chemistry including students and industrialists the text uses exercises tables and figures to help the reader with their understanding of the topic *Students Study Guide* Fredeen, Pearson, 2003-07 **Analytical Chemistry** Frederick Pearson Treadwell, 1928 Chemistry Rothstein, 1995

Mathematical Modelling, Optimization, Analytic and Numerical Solutions Pammy Manchanda, René Pierre Lozi, Abul Hasan Siddiqi, 2020-02-04 This book discusses a variety of topics related to industrial and applied mathematics focusing on wavelet theory sampling theorems inverse problems and their applications partial differential equations as a model of real world problems computational linguistics mathematical models and methods for meteorology earth systems environmental and medical science and the oil industry It features papers presented at the International Conference in Conjunction with 14th Biennial Conference of ISIAM held at Guru Nanak Dev University Amritsar India on 2 4 February 2018 The conference has emerged as an influential forum bringing together prominent academic scientists experts from industry and researchers The topics discussed include Schrodinger operators quantum kinetic equations and their application extensions of fractional integral transforms electrical impedance tomography diffuse optical tomography Galerkin method by using wavelets a Cauchy problem associated with Korteweg de Vries equation and entropy solution for scalar conservation laws This book motivates and inspires young researchers in the fields of industrial and applied mathematics Interactions in Electrolyte Solutions George H. Nancollas, 1966 *Solutions Manual* Joseph Topich, 2000-10 **Solutions Manual to Accompany Organic Chemistry** Jonathan Clayden, Stuart Warren, Stuart G. Warren, 2013 This text contains detailed worked solutions to all the end of chapter exercises in the textbook Organic Chemistry Notes in tinted boxes in the page margins highlight important principles and comments **Introduction to Modern Inorganic Chemistry, 6th edition** R.A. Mackay, W.

Henderson,2017-12-21 This popular and comprehensive textbook provides all the basic information on inorganic chemistry that undergraduates need to know For this sixth edition the contents have undergone a complete revision to reflect progress in areas of research new and modified techniques and their applications and use of software packages Introduction to Modern Inorganic Chemistry begins by explaining the electronic structure and properties of atoms then describes the principles of bonding in diatomic and polyatomic covalent molecules the solid state and solution chemistry Further on in the book the general properties of the periodic table are studied along with specific elements and groups such as hydrogen the s elements the lanthanides the actinides the transition metals and the p block Simple and advanced examples are mixed throughout to increase the depth of students understanding This edition has a completely new layout including revised artwork case study boxes technical notes and examples All of the problems have been revised and extended and include notes to assist with approaches and solutions It is an excellent tool to help students see how inorganic chemistry applies to medicine the environment and biological topics

How to Solve General Chemistry Problems Robert S. Boikess,2009 This is a self teaching text that helps students solve problems found in most general chemistry textbooks Its sequence of chapters is arranged to coincide closely with the sequence in most popular general chemistry textbooks making it easy for today s student to use it as a supplement to any general chemistry textbook

Lectures on Physics Richard P. Feynman,2006

Mechanisms of Reactions of Metal Complexes in Solution Debabrata Banerjee,M K

Bharty,2022-09-21 Reaction Mechanisms of Metal Complexes in Solution provides a comprehensive overview of an often overlooked research area Despite its importance and recent reshaping of the field many inorganic chemists have lost an appreciation for the significance of stability constants and the thermodynamic aspects of complex formation Ideal for newcomers and established researchers in the field this book is a complete treatment of the area covering advanced topics with relevance to biomedical applications extraction metallurgy food chemistry and a wealth of other industrial processes and research areas The book will be of particular interest to postgraduates with an interest in coordination chemistry catalysis supramolecular chemistry metallobiology and related aspects of biochemistry

Science Instruction in the Middle and Secondary Schools Eugene L. Chiappetta,Thomas R. Koballa,2006 For science instruction in middle and secondary schools On Reserve for Edu 427

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 - The Rise of Digital Reading Pearson Chemistry Answer Key Ch 16 Solutions
 - Advantages of eBooks Over Traditional Books
2. Identifying Pearson Chemistry Answer Key Ch 16 Solutions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Pearson Chemistry Answer Key Ch 16 Solutions
 - User-Friendly Interface
4. Exploring eBook Recommendations from Pearson Chemistry Answer Key Ch 16 Solutions
 - Personalized Recommendations
 - Pearson Chemistry Answer Key Ch 16 Solutions User Reviews and Ratings
 - Pearson Chemistry Answer Key Ch 16 Solutions and Bestseller Lists
5. Accessing Pearson Chemistry Answer Key Ch 16 Solutions Free and Paid eBooks
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 - Pearson Chemistry Answer Key Ch 16 Solutions eBook Subscription Services
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 - Highlighting and Note-Taking Pearson Chemistry Answer Key Ch 16 Solutions
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- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Pearson Chemistry Answer Key Ch 16 Solutions
9. Balancing eBooks and Physical Books Pearson Chemistry Answer Key Ch 16 Solutions
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Pearson Chemistry Answer Key Ch 16 Solutions
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Pearson Chemistry Answer Key Ch 16 Solutions
- Setting Reading Goals Pearson Chemistry Answer Key Ch 16 Solutions
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Pearson Chemistry Answer Key Ch 16 Solutions
- Fact-Checking eBook Content of Pearson Chemistry Answer Key Ch 16 Solutions
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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