

CHEMICAL REACTIONS AND EQUATIONS

Objective Questions

Multiple Choice Questions

1. Which of the statements about the reaction below are incorrect ?

$$2\text{PbO(s)} + \text{C(s)} \longrightarrow 2\text{Pb(l)} + \text{CO}_2\text{(g)}$$

- (i) Lead is getting reduced
- (ii) Carbon dioxide is getting oxidised
- (iii) Carbon is getting oxidised
- (iv) Lead oxide is getting reduced
- (a) (i) and (ii)
- (b) (i) and (iii)
- (c) (i), (ii) and (iii)
- (d) all of these.

2. $\text{Fe}_2\text{O}_3 + 2\text{Al} \longrightarrow \text{Al}_2\text{O}_3 + 2\text{Fe}$

The above reaction is an example of a—

(2024)

- (a) combination reaction
- (b) double displacement reaction
- (c) decomposition reaction
- (d) displacement reaction.

3. What happens when dilute hydrochloric acid is added to iron fillings ? (2022)

- (a) Hydrogen gas and iron chloride are produced
- (b) Chlorine gas and iron hydroxide are produced
- (c) No reaction takes place
- (d) Iron salt and water are produced.

In the reaction $2\text{FeCl}_2 + \text{Cl}_2 \longrightarrow 2\text{FeCl}_3$ Chlorine is a—

- (a) Reducing agent
- (b) Oxidizing agent
- (c) Catalyst
- (d) Inert medium.

Rancidity can be prevented in food items by—

- (a) Mixing antioxidant
- (b) Keeping in refrigerator
- (c) Keeping away from light
- (d) All of these.

Which among the following is not an oxidizing agent ?

- (a) Oxygen
- (b) Conc. Sulphuric acid
- (c) Chlorine
- (d) Hydrogen.

Which among the following gases are used to keep the oily food items fresh ?

- (a) Carbon dioxide and Oxygen
- (b) Nitrogen and helium
- (c) Helium and Oxygen
- (d) Nitrogen and Oxygen.

Chemically rust is—

- (a) Hydrated ferrous oxide
- (b) Ferric oxide
- (c) Hydrated ferric oxide
- (d) None of these.

An exothermic reaction in which glucose combines with oxygen in the cells of our body and provides energy—

- (a) Respiration
- (b) Digestion
- (c) Excretion
- (d) None of these.

Pearson Education Chapter 1chemical Reactions Answers

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George**

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Chemistry: The Central Science Theodore L. Brown,H. Eugene LeMay Jr.,Bruce E. Bursten,Catherine Murphy,Patrick Woodward,Steven Langford,Dalius Sagatys,Adrian George,2013-10-04 If you think you know the Brown LeMay Bursten Chemistry text think again In response to market request we have created the third Australian edition of the US bestseller Chemistry The Central Science An extensive revision has taken this text to new heights Triple checked for scientific accuracy and consistency this edition is a more seamless and cohesive product yet retains the clarity innovative pedagogy functional problem solving and visuals of the previous version All artwork and images are now consistent in quality across the entire text And with a more traditional and logical organisation of the Organic Chemistry content this comprehensive text is the source of all the information and practice problems students are likely to need for conceptual understanding development of problem solving skills reference and test preparation **Chemistry insights 'O' level** Rex M. Heyworth,2007

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