

REVIEW OF BASIC CHEMISTRY ANSWER KEY

1. Name the following elements. Spelling counts:

H *hydrogen*
S *sulphur*
Cl *chlorine*
C *carbon*
N *nitrogen*
Na *sodium*
Pb *lead*
Ag *silver*

2. Write the symbols for the following elements.

sodium *Na*
phosphorus *P*
fluorine *F*
magnesium *Mg*
potassium *K*
calcium *Ca*
zinc *Zn*
iron *Fe*

3. Using a periodic table, record the atomic number for the following elements:

gold (Au) **79** copper (Cu) **29**

4. Using a periodic table, record the atomic mass for the following elements to one decimal place:

chlorine (Cl) **35.5** calcium (Ca) **40.1**

5. Complete the following table. Use the information provided in the chart – not a periodic table – to determine atomic masses, but you may refer to a periodic table to name the element.

Element	Atomic Number	Atomic Mass	Protons	Neutrons	Electrons
Lead	82	207	82	125	82
Argon	18	58	18	40	18
Barium	56	137	56	61	56
Tin	50	112	50	62	50

6. Match the unit with its description:

F 1. amount of substance
C 2. density
B 3. mass
D 4. molar mass
A 5. energy
E 6. volume

A. joule
B. g
C. g/mL
D. g/mole
E. L
F. mole

World Of Chemistry Basic Review Sheet

Joacim Rocklöv



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Table of Contents World Of Chemistry Basic Review Sheet

1. Understanding the eBook World Of Chemistry Basic Review Sheet
 - The Rise of Digital Reading World Of Chemistry Basic Review Sheet
 - Advantages of eBooks Over Traditional Books
2. Identifying World Of Chemistry Basic Review Sheet
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an World Of Chemistry Basic Review Sheet
 - User-Friendly Interface
4. Exploring eBook Recommendations from World Of Chemistry Basic Review Sheet
 - Personalized Recommendations
 - World Of Chemistry Basic Review Sheet User Reviews and Ratings
 - World Of Chemistry Basic Review Sheet and Bestseller Lists
5. Accessing World Of Chemistry Basic Review Sheet Free and Paid eBooks
 - World Of Chemistry Basic Review Sheet Public Domain eBooks
 - World Of Chemistry Basic Review Sheet eBook Subscription Services
 - World Of Chemistry Basic Review Sheet Budget-Friendly Options

6. Navigating World Of Chemistry Basic Review Sheet eBook Formats
 - ePub, PDF, MOBI, and More
 - World Of Chemistry Basic Review Sheet Compatibility with Devices
 - World Of Chemistry Basic Review Sheet Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of World Of Chemistry Basic Review Sheet
 - Highlighting and Note-Taking World Of Chemistry Basic Review Sheet
 - Interactive Elements World Of Chemistry Basic Review Sheet
8. Staying Engaged with World Of Chemistry Basic Review Sheet
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers World Of Chemistry Basic Review Sheet
9. Balancing eBooks and Physical Books World Of Chemistry Basic Review Sheet
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection World Of Chemistry Basic Review Sheet
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine World Of Chemistry Basic Review Sheet
 - Setting Reading Goals World Of Chemistry Basic Review Sheet
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of World Of Chemistry Basic Review Sheet
 - Fact-Checking eBook Content of World Of Chemistry Basic Review Sheet
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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