

PRE TEST

Chapter 6 Test-General/Honors Chemistry

Part 1: Matching

Match each description in Column B with the correct term in Column A. Write the letter of the correct description on the line.

- | | | | |
|-------|----------|---------------------------|--|
| C.2.2 | <u>a</u> | 1. cation | a. a positively charged ion |
| C.2.2 | <u>B</u> | 2. electronegativity | b. an element whose highest occupied <i>s</i> or <i>p</i> sublevels are partially filled |
| C.2.5 | <u>J</u> | 3. inner transition metal | c. the energy required to remove an electron from an atom |
| C.2.6 | <u>I</u> | 4. alkali earth metal | d. good conductor of heat and electric current |
| C.2.6 | <u>h</u> | 5. noble gas | e. any element whose highest occupied <i>s</i> sublevel and a nearby <i>d</i> sublevel contain electrons |
| C.2.5 | <u>b</u> | 6. representative element | f. any nonmetal in group 7A |
| C.2.2 | <u>d</u> | 7. metal | g. the tendency of an atom to attract electrons when it's in a compound (how much an atom wants an electron) |
| C.2.6 | <u>f</u> | 8. halogen | h. an element in which the highest occupied <i>s</i> and <i>p</i> sublevels are completely filled |
| C.2.5 | <u>e</u> | 9. transition metal | i. an element in group 2A |
| C.2.2 | <u>c</u> | 10. ionization energy | j. an element whose highest <i>s</i> sublevel and a nearby <i>f</i> sublevel contain electrons |

Part 2: Multiple Choice

Choose the best answer and write its letter on the line.

- | | | | | |
|-------|----------|--------------------------------------|--------------------------|------------------------------|
| C.2.6 | <u>b</u> | 11. The alkali metals do not include | a. Li | c. Na |
| | | | b. Ca | d. Rb |
| C.2.6 | <u>c</u> | 12. The element iodine, I, is a | a. period 5 alkali metal | c. period 5 halogen |
| | | | b. period 4 halogen | d. period 5 transition metal |

World Of Chemistry Chapter Test Answers

William H. Brown, Thomas Poon



World Of Chemistry Chapter Test Answers:

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Gaither's Dictionary of Scientific Quotations Carl C. Gaither, Alma E. Cavazos-Gaither, 2008-01-08 Scientists and other keen observers of the natural world sometimes make or write a statement pertaining to scientific activity that is destined to live on beyond the brief period of time for which it was intended. This book serves as a collection of these statements from great philosophers and thought influencers of science past and present. It allows the reader quickly to find relevant quotations or citations. Organized thematically and indexed alphabetically by author, this work makes readily available an unprecedented collection of approximately 18 000 quotations related to a broad range of scientific topics.

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