

Chemistry Chapter 8 Test

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- ____ 1. What type of bond would you expect in a compound of oxygen and sodium?
☒ a. ionic
 b. metallic
 c. covalent
 d. polar covalent
- ____ 2. The symbols δ^- and δ^+ indicate which of the following?
 a. electronegativity
 b. electrostatic force
 c. charge
☒ d. partial charge
- ____ 3. Which of the following represents a nonpolar covalent molecule?
 a. NaCl
☒ b. Br₂
 c. HCN
 d. Au
- ____ 4. The measure of an atom's ability to attract electrons to itself in a chemical bond is called _____.
 a. electron affinity
 b. polarity
☒ c. electronegativity
 d. ionization energy
- ____ 5. The polarity of a bond depends on the difference in the atoms' _____.
 a. electron affinities
☒ b. electronegativities
 c. atomic radii
 d. ionization energies
- ____ 6. What type of bond would you expect to form between nitrogen and potassium?
☒ a. ionic
 b. covalent
 c. polar covalent
 d. metallic
- ____  7. High electronegativity differences are found in _____ bonds.
 a. metallic
☒ b. ionic metal - nonmetal
 c. covalent
 d. Pauling
- ____  8. To achieve an octet when combining with potassium, would an atom of a Group 16 element be expected to lose, gain, or share electrons?
 a. lose
☒ b. gain
 c. share
☒ d. This cannot be determined from the given information. x
- ____ 9. Which of the following does not exist as a diatomic molecule?
☒ a. argon
 b. hydrogen
 c. chlorine
 d. oxygen
 H N O F Cl Br I
 Gas : H₂ N₂ O₂ F₂ Cl₂ Br₂ I₂
- ____ 10. Which of these is not a covalent molecule?
 a. F₂ =
 b. CO =
☒ c. LiF
 d. NH₃
- ____ 11. A dash in a Lewis structure represents _____.
☒ a. one electron
 b. two electrons
 c. four electrons
 d. unpaired electrons
- ____ 12. The central atom in a Lewis structure is usually the atom _____.

World Of Chemistry Test Answers

Robyn V. Young



World Of Chemistry Test Answers:

Survey of Science History & Concepts Parent Lesson Plan, 2013-08-01 Survey of Science History Concepts Course Description Students will study four areas of science Scientific Mathematics Physics Biology and Chemistry Students will gain an appreciation for how each subject has affected our lives and for the people God revealed wisdom to as they sought to understand Creation Each content area is thoroughly explored giving students a good foundation in each discipline Semester 1 Math and Physics Numbers surround us Just try to make it through a day without using any It s impossible telephone numbers calendars volume settings shoe sizes speed limits weights street numbers microwave timers TV channels and the list goes on and on The many advancements and branches of mathematics were developed through the centuries as people encountered problems and relied upon math to solve them It s amazing how ten simple digits can be used in an endless number of ways to benefit man The development of these ten digits and their many uses is the fascinating story in Exploring the World of Mathematics Physics is a branch of science that many people consider to be too complicated to understand John Hudson Tiner puts this myth to rest as he explains the fascinating world of physics in a way that students can comprehend Did you know that a feather and a lump of lead will fall at the same rate in a vacuum Learn about the history of physics from Aristotle to Galileo to Isaac Newton to the latest advances Discover how the laws of motion and gravity affect everything from the normal activities of everyday life to launching rockets into space Learn about the effects of inertia first hand during fun and informative experiments Exploring the World of Physics is a great tool for student who want to have a deeper understanding of the important and interesting ways that physics affects our lives Semester 2 Biology and Chemistry The field of biology focuses on living things from the smallest microscopic protozoa to the largest mammal In this book you will read and explore the life of plants insects spiders and other arachnids life in water reptiles birds and mammals highlighting God s amazing creation You will learn about biological classification how seeds spread around the world long term storage of energy how biologists learned how the stomach digested food the plant that gave George de Mestral the idea of Velcro and so much more For most of history biologists used the visible appearance of plants or animals to classify them They grouped plants or animals with similar looking features into families Starting in the 1990 s biologists have extracted DNA and RNA from cells as a guide to how plants or animals should be grouped Like visual structures these reveal the underlying design of creation Exploring the World of Biology is a fascinating look at life from the smallest proteins and spores to the complex life systems of humans and animals Chemistry is an amazing branch of science that affects us every day yet few people realize it or even give it much thought Without chemistry there would be nothing made of plastic there would be no rubber tires no tin cans no televisions no microwave ovens or something as simple as wax paper This book presents an exciting and intriguing tour through the realm of chemistry as each chapter unfolds with facts and stories about the discoveries of discoverers Find out why pure gold is not used for jewelry or coins Join Humphry Davy as he made many chemical discoveries and learn how

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