

Thermochemistry

Multiple Choice Questions

- B 1. When the three types of energy changes are arranged in order of decreasing molar heat values, what is the correct order?
A. chemical, phase, nuclear
B. nuclear, chemical, phase
C. nuclear, phase, chemical
D. phase, chemical, nuclear
- B 2. The temperature remains constant as energy is added to a substance. How may the substance be changing?
A. from a gas to a solid
B. from a liquid to a gas
C. from a liquid to a solid
D. in the amount of kinetic energy
- A 3. Which metal requires the most energy to raise 1.00 g of it by 1.00°C?
A. aluminum
B. copper
C. lead
D. nickel
- B 4. A substance has a molar heat of combustion of -850.4 kJ/mol. When 0.285 mol of the substance is burned in a calorimeter containing 8.60 kg of water, what is the increase of the water temperature?
A. 0.156°C
B. 6.41°C
C. 7.89°C
D. 12.8°C
- B 5. The addition of 9.54 kJ of heat is required to raise the temperature of 225.0 g of a liquid hydrocarbon from 20.5°C to 45.0°C. What is the heat capacity of this hydrocarbon?
A. 0.94 J/g°C
B. 1.73 J/g°C
C. 1.88 J/g°C
D. 9.42 J/g°C
- C 6. If the molar heat of fusion of sodium is 2.63 kJ/mol, how much energy is needed to melt 180.0 g of solid sodium at its melting point?
A. 2.63 kJ
B. 2.93 kJ
C. 20.6 kJ
D. 473 kJ
- B 7. Which process is exothermic?
A. the boiling of liquid nitrogen
B. the freezing of water
C. the sublimation of dry ice
D. the vaporization of water
- A 8. If the heat of formation for the chemical reaction, $S_8(s) + 12O_2(g) \rightarrow 8SO_2(g)$, is -395.7 kJ/mol of SO_2 , how can the equation be rewritten to include the heat term?
A. $S_8(s) + 12O_2(g) \rightarrow 8SO_2(g) + 3166 \text{ kJ}$
B. $S_8(s) + 12O_2(g) \rightarrow 8SO_2(g) - 395.7 \text{ kJ}$
C. $S_8(s) + 12O_2(g) \rightarrow 8SO_2(g) + 395.7 \text{ kJ}$
D. $S_8(s) + 12O_2(g) + 3166 \text{ kJ} \rightarrow 8SO_2(g)$
- A 9. Which statement correctly describes the energy changes that take place when a solid changes to a liquid at constant temperature?
A. The potential energy increases, and the kinetic energy remains constant.
B. The potential energy remains constant, and the kinetic energy increases.
C. The potential energy decreases, and the kinetic energy decreases.
D. The potential energy increases, and the kinetic energy decreases.
- B 10. Why does warm water have a more rapid rate of evaporation than cold water?
A. It has a higher viscosity.
B. Its molecules have a higher average kinetic energy.
C. Its molecules have more attraction for one another.
D. More of its molecules have nearly the same kinetic energy.
- B 11. If the heat of fusion of a substance is 20 kJ/mol, what heat is released when 1.0 mol of liquid at the melting point freezes?
A. 10 kJ
B. 20 kJ
C. 40 kJ
D. 60 kJ
- C 12. The amount of 2.00 mol of a solid ionic compound was dissolved in 225 mL of water in an insulated container. The initial temperature of the water was 20.0°C and the final temperature of the water was 47.8°C. What is ΔH_{soln} ?
A. -27.8 kJ/mol
B. -26.2 kJ/mol
C. -13.1 kJ/mol
D. -6.26 kJ/mol
- D 13. What does it mean if the ΔH value for a chemical reaction is positive?
A. kinetic energy is increasing in the system
B. potential energy is decreasing in the system
C. products have less potential energy than reactants
D. reactants have less potential energy than products.
- A 14. Living plants produce glucose in the process of photosynthesis according to this equation:
 $6CO_2(g) + 6H_2O(l) + \text{energy} \rightarrow C_6H_{12}O_6(s) + 6O_2(g)$
Is this reaction endothermic or exothermic, and is the value of ΔH° positive or negative?
A. endothermic, positive
B. endothermic, negative
C. exothermic, positive
D. exothermic, negative

Real Life Thermochemistry Questions

Mocktime Publication



Real Life Thermochemistry Questions:

Proceedings of the 1st Lawang Sewu International Symposium on Humanities and Social Sciences 2022

(LEWIS 2022) Dodi Mulyadi, Siti Aimah, Eny Winaryati, 2023-06-23 This is an open access book 1st Lawang Sewu International Symposium 2022 on Humanities and Social Sciences is an annual international symposium held by Universitas Muhammadiyah Semarang Symposium will be held on November 29 2022 in Semarang Central Java Indonesia by online Lecturers professionals researches and students are invited in 1st Lawang Sewu International Symposium 2022 on Humanities and Social Sciences Multi field study including Education Psychology Economics and management are welcome The submitted papers must meet the criteria including originality novelty not yet published and must be written in English language Symposium will be held through online due to Covid 19 pandemic situation The 9th Annual International Seminar on Trends in Science and Science Education (AISTSSE) 2022 ,2023-10-04 This is the ninth time we are hosting this seminar and we are proud to inform you that this seminar is an annual event in our calendar and has been held every year since 2014 This year for the third year we are holding it via Zoom meeting online meeting due to Covid 19 pandemic We are inviting internationally recognized speakers from several countries to share their latest discoveries in the fields of Biology Chemistry Physics Mathematics and Science Education Well known researchers in science and science education will share their experiences and knowledge so that we can stay up to date with the latest information This is one of the goals of this seminar As science researchers we realize the importance of information exchange among us The new information enlightens our minds and gives us ideas on what to do next in our research and how to do it This new information often becomes the foundation for our next project in particular and sets the research trends for the upcoming year in general Information exchange also keeps us updated allowing us to give and receive suggestions and critiques that will lead to better results Therefore we need a forum where we can share and exchange information Seminars conferences and other scientific gatherings are the media through which we can do this Organizer Faculty of Mathematics and Natural Sciences of Universitas Negeri Medan Where Web Seminar via Zoom Meeting When Tuesday 8th November 2022 Theme The development of industrial based research in science and science education to improve research innovation strategy Topics AISTSSE 2020 included following topics 1 Mathematics Science 2 Mathematics Education 3 Physics Science 4 Physics Education 5 Biology Science 6 Biology Education 7 Chemistry Science 8 Chemistry Education 9 Computer Science 10 Science Education Scientific Committee 1 Prof Dr Syawal Gultom M Pd Universitas Negeri Medan Indonesia 2 Prof Dr Marleen Kamperman University of Groningen Netherland 3 Prof Manihar Situmorang M Sc Ph D Universitas Negeri Medan Indonesia 4 Prof Tsunenori Mine School of Engineering Department of Electrical Engineering and Computer Science Kyushu University Japan 5 Prof Dian Armanto M Pd Universitas Negeri Medan Indonesia 6 Prof Dr Herbert Sipahutar M Sc Universitas Negeri Medan Indonesia 7 Prof Abedel Karrem Nasser M Alomari Department of Mathematics Faculty of Science

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Table of Contents Real Life Thermochemistry Questions

1. Understanding the eBook Real Life Thermochemistry Questions
 - The Rise of Digital Reading Real Life Thermochemistry Questions
 - Advantages of eBooks Over Traditional Books
2. Identifying Real Life Thermochemistry Questions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in a Real Life Thermochemistry Questions
 - User-Friendly Interface
4. Exploring eBook Recommendations from Real Life Thermochemistry Questions
 - Personalized Recommendations
 - Real Life Thermochemistry Questions User Reviews and Ratings
 - Real Life Thermochemistry Questions and Bestseller Lists
5. Accessing Real Life Thermochemistry Questions Free and Paid eBooks
 - Real Life Thermochemistry Questions Public Domain eBooks
 - Real Life Thermochemistry Questions eBook Subscription Services
 - Real Life Thermochemistry Questions Budget-Friendly Options

6. Navigating Real Life Thermochemistry Questions eBook Formats
 - ePub, PDF, MOBI, and More
 - Real Life Thermochemistry Questions Compatibility with Devices
 - Real Life Thermochemistry Questions Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Real Life Thermochemistry Questions
 - Highlighting and Note-Taking Real Life Thermochemistry Questions
 - Interactive Elements Real Life Thermochemistry Questions
8. Staying Engaged with Real Life Thermochemistry Questions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Real Life Thermochemistry Questions
9. Balancing eBooks and Physical Books Real Life Thermochemistry Questions
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Real Life Thermochemistry Questions
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Real Life Thermochemistry Questions
 - Setting Reading Goals Real Life Thermochemistry Questions
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Real Life Thermochemistry Questions
 - Fact-Checking eBook Content of Real Life Thermochemistry Questions
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

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