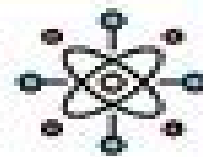


# CHEMICAL BONDS



## IONIC, COVALENT, & METALLIC

**FORMS OF CHEMICAL BONDS**  
Bonding: combination of atoms

Atoms combine to form molecules or compounds by sharing or transferring electrons. This process is called bonding. The resulting molecule or compound has properties different from the individual atoms.

**Types of Bonds:**

- Covalent Bond:** Formed by sharing electrons between two atoms. Example:  $H_2$ ,  $H_2O$ ,  $CO_2$ .
- Ionic Bond:** Formed by the transfer of electrons from one atom to another, creating oppositely charged ions that attract each other. Example:  $NaCl$ ,  $MgO$ .
- Metallic Bond:** Formed by the attraction between positive metal ions and a sea of delocalized electrons. Example:  $Fe$ ,  $Cu$ ,  $Al$ .

**Covalent, Ionic, and Metallic Bonding**

Covalent bonding occurs when two atoms share one or more pairs of electrons. This results in a stable molecule. Ionic bonding occurs when one atom transfers an electron to another atom, creating a positive ion and a negative ion. These ions are attracted to each other. Metallic bonding occurs in metals, where the atoms are arranged in a regular lattice and the outer electrons are delocalized, forming a 'sea of electrons'.

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**Types of Bonds:**

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## GUIDED READING ACTIVITY

# Reading Comprehension On Ionic And Covalent Bonds For Middle School

**Jiyuan Zhang**



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