

- c. C) Italy
 - d. D) Ottoman Empire
10. The United States entered World War I in the year:
- a. A) 1914
 - b. B) 1915
 - c. C) 1916
 - d. D) 1917
11. What was the main purpose of the League of Nations?
- a. A) To prevent future wars
 - b. B) To promote trade between nations
 - c. C) To spread communism
 - d. D) To create a world government
12. Which battle is known for being one of the longest and bloodiest engagements of World War I?
- a. A) The Battle of the Marne
 - b. B) The Battle of Verdun
 - c. C) The Battle of Gallipoli
 - d. D) The Battle of Tannenberg
13. The Treaty of Versailles was particularly harsh towards which nation?
- a. A) Austria
 - b. B) Germany
 - c. C) Russia
 - d. D) France
14. What did the term "total war" refer to in the context of World War I?
- a. A) A conflict where only professional soldiers fought
 - b. B) A war that involved only naval battles
 - c. C) A war that required the mobilization of each nation's entire resources
 - d. D) A war where all nations agreed on the total number of troops to be involved
15. Which of the following was not a consequence of World War I?
- a. A) The collapse of the Ottoman Empire
 - b. B) The start of World War II
 - c. C) The Russian Revolution
 - d. D) The establishment of the European Union
16. The sinking of the Lusitania escalated tensions between Germany and:
- a. A) France
 - b. B) The United States
 - c. C) Russia
 - d. D) Britain
17. The concept of self-determination was promoted by which U.S. President during World War I?
- a. A) Franklin D. Roosevelt
 - b. B) Woodrow Wilson
 - c. C) Theodore Roosevelt
 - d. D) Abraham Lincoln
18. Which of these countries switched sides in the middle of World War I?

World History I Testbank

**Bonnie G. Smith, Marc Van De
Mieroop, Richard von Glahn, Kris Lane**

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for determining the limiting reactant and the amount of excess reactant in a chemical reaction

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web 33mol O_2 6mol Br_2 11mol O_2 18mol Br_2 identify the limiting reactant s and excess reactant s the limiting reactant is O_2 since it would yield the least amount of product 18 mol Br_2 the excess reactant is $\text{C}_2\text{H}_3\text{Br}_3$ since its complete reaction would have yielded up to 54 mol Br_2 think about your result

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web step 4 the reactant that produces a smaller amount of product is the limiting reagent mg mg produces less mgo mgo than does O_2 0.2 3 98 g mgo vs 25.2 g mgo therefore mg is the limiting reagent in this reaction step 5 the reactant that produces a larger amount of product is the excess reagent

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web if you start with 14.82 g of Ca(OH)_2 c a o h 2 and 16.35 g of H_2SO_4 h 2 s o 4 a determine the limiting reagent b determine the number of moles of H_2O h 2 o produced c determine the number of grams of CaSO_4 c a s o 4 produced d determine the number of grams of excess reagent left 1 make sure the equation is balanced

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