PRE TEST

Chapter 6 Test-General/Honors Chemistry

Part 1: Matching

Match each description in Column B with the correct term in Column A. Write the letter of the correct description on the line.

C.2.2 ______1. cation a. a positively charged ion C2.2 8 2. electronegativity b. an element whose highest occupied s or p sublevels are partially filled C.2.5 3. inner transition metal c. the energy required to remove an electron from an atom C.2.6 4. alkali earth metal d. good conductor of heat and electric current. C.2.6 h 5. noble gas e. any element whose highest occupied s sublevel and a nearby d sublevel contain electrons C.2.5 6. representative element f. any nonmetal in group 7A c 2 2 ______ 7. metal g. the tendency of an atom to attract electrons when it's in a compound (how much an atom wants an electron) C.2.6 f 8. halogen h. an element in which the highest occupied s and p sublevels are completely filled C2.5 ______9. transition metal i. an element in group 2A C.2.2 c 10. ionization energy an element whose highest s sublevel and a nearby f sublevel contain electrons

Part 2: Multiple Choice

Choose the best answer and write its letter on the line.

- C.2.6 __b __11. The alkali metals do not include
 a. l.i c. Na
 b. Ca d. Rb
- C.2.6 ____12. The element iodine, I, is a a. period S alkali metal b. period 4 halogen

c. period 5 halogen d. period 5 transition metal

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Table of Contents World Of Chemistry Chapter Test

- 1. Understanding the eBook World Of Chemistry Chapter Test
 - The Rise of Digital Reading World Of Chemistry Chapter Test
 - Advantages of eBooks Over Traditional Books
- 2. Identifying World Of Chemistry Chapter Test
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an World Of Chemistry Chapter Test
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from World Of Chemistry Chapter Test
 - Personalized Recommendations
 - World Of Chemistry Chapter Test User Reviews and Ratings

- World Of Chemistry Chapter Test and Bestseller Lists
- 5. Accessing World Of Chemistry Chapter Test Free and Paid eBooks
 - World Of Chemistry Chapter Test Public Domain eBooks
 - World Of Chemistry Chapter Test eBook Subscription Services
 - World Of Chemistry Chapter Test Budget-Friendly Options
- 6. Navigating World Of Chemistry Chapter Test eBook Formats
 - o ePub, PDF, MOBI, and More
 - World Of Chemistry Chapter Test Compatibility with Devices
 - World Of Chemistry Chapter Test Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of World Of Chemistry Chapter Test
 - Highlighting and Note-Taking World Of Chemistry Chapter Test
 - Interactive Elements World Of Chemistry Chapter Test
- 8. Staying Engaged with World Of Chemistry Chapter Test
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers World Of Chemistry Chapter Test
- 9. Balancing eBooks and Physical Books World Of Chemistry Chapter Test
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection World Of Chemistry Chapter Test
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine World Of Chemistry Chapter Test
 - Setting Reading Goals World Of Chemistry Chapter Test
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of World Of Chemistry Chapter Test
 - Fact-Checking eBook Content of World Of Chemistry Chapter Test
 - Distinguishing Credible Sources

- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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