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web may 15 2020 toluene 1 oxidation with kmno4 mechanism 2 benzoic acid c6h5cooh socl2 3 benzoyl chloride c6h5cocl lialh oc ch3 3 h 4 benzaldehyde c6h5cho ch2ohch2oh h step 1 toluene oxidation with kmno4 kmno4 is a strong oxidising agent which oxidise toluene to benzaldehyde in 1st step

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web oct 15 2014 $\,$ 1 answer sorted by 6 here are the three equations describing the reduction of manganese and concurrent oxidation of whatever substrate may be present under basic neutral and acidic conditions respectively mnx 7ox4x ex mnx 6ox4x2 basic m n x 7 o x 4 x e x m n x 6 o x 4 x 2 b a s i c

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web mar $15\ 2019$ what is the mechanism of toluene oxidation by kmno4 it is quite simple reaction that kmno4 can oxidize benzylic hydrogen in toluene and synthesize benzoic acid however i cannot find exact mechanism of reaction in

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web description treatment of an alkylbenzene with potassium permanganate results in oxidation to give the benzoic acid notes the position directly adjacent to an aromatic group is called the benzylic position the reaction only works if there is a hydrogen attached to the carbon examples

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web may 1 2004 the oxidation of toluene by aq kmno 4 under hydrodynamic cavitation was taken as a model reaction and various parameters have been optimized the oxidation of toluene by aq kmno 4 gives benzoic acid scheme 1 kmno 4 in turn is reduced to mno 2 the reaction did not give any other byproduct