



**PHYSICAL SCIENCES (CHEMISTRY)  
GRADE 11 TERM 3 FORMAL EXPERIMENT - PART 2**

**TIME: 1 HOUR**

**VERIFICATION OF BOYLE'S LAW  
INDIVIDUAL WORK (should be done under strict supervision).**

**1. Write a scientific report which must include the following:**

**1.1 Aim**

**(1)**

**1.2 Apparatus and method**

**(5)**

**2. Observations and Recording of results** (Copy the table below onto your book or sheet of paper and transfer the results you recorded when conducting the experiment). Calculate  $pV$  for each set of results and record your answers in the table.

| Volume (V) $\text{cm}^3$ | Pressure (p) kPa |  | $pV$ (J) |
|--------------------------|------------------|--|----------|
|                          |                  |  |          |
|                          |                  |  |          |
|                          |                  |  |          |
|                          |                  |  |          |
|                          |                  |  |          |
|                          |                  |  |          |

**(6)**

**3. Analysis of the results:**

**3.1 Identify the following variables for the experiment:**

- Independent
- Dependent
- Controlled

**(3)**

**3.2 Use the graph paper provided by your teacher to draw graphs of V versus p and V versus  $\frac{1}{p}$ .**

**(18)**

**4. Conclusion**

**4.1 Examine the results in your table as well as the two graphs you have drawn, what happens to the volume of the gas in the tube as the pressure is decreased.**

**(1)**

**4.2 Write down a conclusion for this experiment.**

**(3)**

**5**

**Questions and discussion:**

**5.1 Explain why you have to wait for a few minutes before you can take readings.**

**(2)**

**5.2 Explain the relationship between pressure and volume in terms of the kinetic theory of gases.**

**(4)**

# Nkangala District Question Paper For Grade 11 Physical Science First Test 2014

**E Durkheim**



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## **Table of Contents Nkangala District Question Paper For Grade 11 Physical Science First Test 2014**

1. Understanding the eBook Nkangala District Question Paper For Grade 11 Physical Science First Test 2014
  - The Rise of Digital Reading Nkangala District Question Paper For Grade 11 Physical Science First Test 2014
  - Advantages of eBooks Over Traditional Books
2. Identifying Nkangala District Question Paper For Grade 11 Physical Science First Test 2014
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Nkangala District Question Paper For Grade 11 Physical Science First Test 2014
  - User-Friendly Interface
4. Exploring eBook Recommendations from Nkangala District Question Paper For Grade 11 Physical Science First Test 2014

- Personalized Recommendations
  - Nkangala District Question Paper For Grade 11 Physical Science First Test 2014 User Reviews and Ratings
  - Nkangala District Question Paper For Grade 11 Physical Science First Test 2014 and Bestseller Lists
5. Accessing Nkangala District Question Paper For Grade 11 Physical Science First Test 2014 Free and Paid eBooks
    - Nkangala District Question Paper For Grade 11 Physical Science First Test 2014 Public Domain eBooks
    - Nkangala District Question Paper For Grade 11 Physical Science First Test 2014 eBook Subscription Services
    - Nkangala District Question Paper For Grade 11 Physical Science First Test 2014 Budget-Friendly Options
  6. Navigating Nkangala District Question Paper For Grade 11 Physical Science First Test 2014 eBook Formats
    - ePub, PDF, MOBI, and More
    - Nkangala District Question Paper For Grade 11 Physical Science First Test 2014 Compatibility with Devices
    - Nkangala District Question Paper For Grade 11 Physical Science First Test 2014 Enhanced eBook Features
  7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Nkangala District Question Paper For Grade 11 Physical Science First Test 2014
    - Highlighting and Note-Taking Nkangala District Question Paper For Grade 11 Physical Science First Test 2014
    - Interactive Elements Nkangala District Question Paper For Grade 11 Physical Science First Test 2014
  8. Staying Engaged with Nkangala District Question Paper For Grade 11 Physical Science First Test 2014
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Nkangala District Question Paper For Grade 11 Physical Science First Test 2014
  9. Balancing eBooks and Physical Books Nkangala District Question Paper For Grade 11 Physical Science First Test 2014
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Nkangala District Question Paper For Grade 11 Physical Science First Test 2014
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Nkangala District Question Paper For Grade 11 Physical Science First Test 2014

- Setting Reading Goals Nkangala District Question Paper For Grade 11 Physical Science First Test 2014
- Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Nkangala District Question Paper For Grade 11 Physical Science First Test 2014
  - Fact-Checking eBook Content of Nkangala District Question Paper For Grade 11 Physical Science First Test 2014
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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