# MULTIPLE CHOICE QUESTIONS

# Oxidation Number



- (A) loss of electrons
- (B) gain of electrons
- (C) increase in the negative valency
- (D) decrease in positive valency.
- When zinc is added to CuSO<sub>4</sub> solution, copper is precipitated, it is so because of
- (A) Reduction of Zn
- (B) Reduction of Cu2+
- (C) Hydrolysis of CuSO4
- (D) Reduction of SO<sub>4</sub><sup>2</sup>.
- The process in which oxidation number increases
  - (A) Reduction
- (B) Hydrolysis
- (C) Oxidation
- (D) Decomposition.
- When zinc metal is added to dilute H2SO4 solution. hydrogen is evolved. In this zinc undergoes
  - (A) Reduction
- (B) Oxidation
- (C) Dissolution
- (D) None of the above.
- In which of the following, oxidation number of chlorine is +5?
  - (A) Cl<sub>2</sub>O<sub>7</sub>

(B) ClO<sub>3</sub>

(C) CIO-

- (D) ClO7.
- 6. Carbon has zero oxidation number in
  - (A) CH4

(B) CH<sub>3</sub>Cl

(C) CCL

- (D) CH<sub>2</sub>Cl<sub>2</sub>.
- 7. The oxidation number of carbon in C12H22O11 is
  - (A) 0

(B) + 22

(C) + 6

- (D) 6.
- 8. In the following

$$3Br_2 + 6CO_3^{2-} + 3H_2O \longrightarrow$$

$$5Br^{-} + BrO_{3}^{-} + 6HCO_{3}^{-}$$

- (A) Bromine is oxidised and carbonate is reduced
- (B) Bromine is reduced and carbonate is oxidised

- (C) Bromine is neither reduced nor oxidised
- (D) Bromine is reduced and oxidised.
- The oxidation number of C in CH4, CH3Cl, CH<sub>2</sub>Cl<sub>2</sub>, CHCl<sub>3</sub> and CCl<sub>4</sub> is respectively
  - (A) 4, -2, 0, +2, +4
  - (B) + 2, 4, 0, -2, -4
  - (C) 4, 2, 0, -2, -4
  - (D) 0, 2, -2, 4, 4,
- Chlorine is in + 3 oxidation state in
  - (A) HCI

(B) HClO<sub>4</sub>

(C) ICI

- (D) CIF<sub>3</sub>.
- Oxidation number of Mn in K<sub>2</sub>MnO<sub>4</sub> is
  - (A) 2

(B) 4

(C) 6

- (D) 7.
- 12. A student states that burning of lime in air is an oxidation process. The reason he gives is that an oxide of the metal is produced on burning. Which one is correct?
  - (A) The statement and reason are true, the reason is correct explanation
  - (B) The statement and reason are true, the explanation is not correct
  - (C) The statement is true but the reason is false
  - (D) The statement is false but the reason is true.
- For the following reaction in the acidic solution

$$MnO_4^- + 8H^+ + 5e^- \longrightarrow Mn^{2+} + 4H_2O_*$$

which of the following gives the true oxidation numbers of the manganese on each side of the equation?

- (A) + 7 to + 6
- (B) + 7 to + 2
- (C) + 4 to + 2
- (D) -1 to +2.
- 14. In which of the following compounds, iron has the lowest oxidation number?
  - (A) Fe(CO)s
  - (B) Fe<sub>2</sub>O<sub>3</sub>
  - (C) K<sub>4</sub> [Fe(CN)<sub>6</sub>]
  - (D) FeSO<sub>4</sub>(NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>6H<sub>2</sub>O.
- 15. The oxidation number and covalency of sulphuin sulphur molecule (S<sub>8</sub>) are

#### Answers

- 1. (A) 11. (0)
  - 2. (B) 12. (D)
- 13. (B)
- 3. (0)
- - 4. (B) 14. (A)
- 5. (B) -
- 6. (D)
- 7. (A)
- 8. (D)
- 9. (A)
- 10. (D)

# **Redox Multiple Choice Questions And Answers**

Barron's Educational Series, Neil D. Jespersen, Pamela Kerrigan

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