Practice Problems: Redox Reactions (Answer Key)

1. Determine the oxidation number of the elements in each of the following compounds:

```
a H<sub>2</sub>CO<sub>3</sub>
H: +1, O: -2, C: +4
b N<sub>2</sub>
N: 0
c Zn(OH)<sub>4</sub><sup>2</sup>·
Zn: 2+, H: +1, O: -2
d NO<sub>2</sub>·
N: +3, O: -2
e LiH
Li: +1, H: -1
f Fe<sub>3</sub>O<sub>4</sub>
Fe: +8/3, O: -2
```

2. Identify the species being oxidized and reduced in each of the following reactions:

 Would you use an oxidizing agent or reducing agent in order for the following reactions to occur?

Redox Reaction Problems With Answer Key

Thomas Bond, Chris Hughes

Redox Reaction Problems With Answer Key:

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