Practice Problems 20.1 Redox Reactions

 Determine the oxidation number of various atoms in a compound, identify what has been oxidized/reduced in a chemical reaction and identify oxidizing/reducing agents

$$Cr_2O_7^{-2}(aq) + Cl^{-}(aq) \rightarrow Cr^{+3}(aq) + Cl_2^{-0}(g)$$

Oxidized: Cl⁻ (aq)

Reduced: Cr2O7-2(aq)

Oxidizing Agent: Cr₂O₇-2(aq)

Reducing Agent: Cl (aq)

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