

Redox Exercise (Part 2)

Chapter 20 Worksheet: Redox ANSWERS

1. Determine what is oxidized and what is reduced in each reaction. Identify the oxidizing agent and the reducing agent, also.



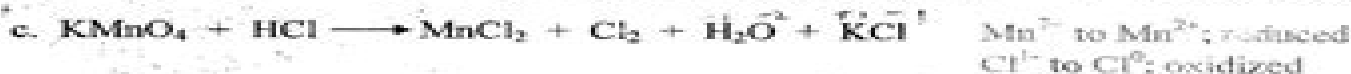
11. Give the oxidation number of each kind of atom or ion.

a. sulfate	b. Sn	c. S^{2-}	d. Fe^{3+}	e. Sn^{4+}	f. nitrate	g. ammonium
2-	0	2-	3+	4+	1-	1+

12. Calculate the oxidation number of chromium in each of the following.

a. Cr_2O_3	b. $\text{Na}_2\text{Cr}_2\text{O}_7$	c. CrSO_4	d. chromate	e. dichromate
3+	6+	2+	6+	6+

13. Use the changes in oxidation numbers to determine which elements are oxidized and which are reduced in these reactions. (Note: it is not necessary to use balanced equations)



14. For each reaction in problem 13, identify the oxidizing agent and reducing agent.

- | | |
|-------------------------------|--------------------------|
| a. oxidizing agent: sulfur | reducing agent: carbon |
| b. oxidizing agent: nitrogen | reducing agent: iodine |
| c. oxidizing agent: manganese | reducing agent: chloride |
| d. oxidizing agent: nitrogen | reducing agent: antimony |

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Holt Chemistry Ralph Thomas Myers, 2004 Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 **National 5 Chemistry with Answers, Second Edition** Barry McBride, Stephen Jeffrey, John Anderson, Paul McCranor, Fran Macdonald, 2018-07-30 Exam Board SQA Level National 5 Subject Chemistry First Teaching September 2017 First Exam Summer 2018 The second edition of this textbook has been fully revised and updated to reflect changes made to the SQA syllabus from 2017 onwards New features include Refreshed content Additional candidate advice Model answers for open ended questions **Holt Chemistry** Holt Rinehart & Winston, 2003-01-24 **Holt Chemistry File**, 1998 This reference is a must for students who need extra help reteaching or extra practice The guide moves students through the same concepts as the text but at a slower pace More descriptive detail along with visual algorithms provides a more structured approach Each chapter closes with a large bank of practice problems Book jacket *Fundamentals of Chemistry: A Modern Introduction (1966)* Frank Brescia, 2012-12-02 Fundamentals of Chemistry A Modern Introduction focuses on the formulas processes and methodologies used in the study of chemistry The book first looks at general and historical remarks definitions of chemical terms and the classification of matter and states of aggregation The text then discusses gases Ideal gases pressure of a gas confined by a liquid Avogadro's Law and Graham's Law are described The book also discusses aggregated states of matter atoms and molecules chemical equations and arithmetic thermochemistry and chemical periodicity The text also highlights the electronic structures of atoms Quantization of electricity spectra of elements quantization of the energy of an electron associated with nucleus the Rutherford Bohr nuclear theory hydrogen atom and representation of the shapes of atomic orbitals are explained The text also highlights the types of chemical bonds hydrocarbons and their derivatives intermolecular forces solutions and chemical equilibrium The book focuses as well on ionic solutions galvanic cells and acids and bases It also discusses the structure and basicity of hydrides and oxides The reactivity of hydrides charge of dispersal and basicity effect of anionic charge inductive effect and basicity and preparation of acids are described The book is a good source of information for readers wanting to study chemistry Satya Prakash's Modern Inorganic Chemistry R D Madan, 2019 Satya Prakash's Modern Inorganic Chemistry is a treatise on the chemistry of elements on the basis of latest theories of Chemistry Initial chapters are devoted to the study of fundamentals of Chemistry such as structure of atom periodic classification of elements chemical bonding and radioactivity to name a few It further graduates to complex discussions not only on extraction properties and uses of the elements but also on preparation properties uses and structure of their important compounds Chemistry of elements and their compounds have been explained on the basis of their position in the long form of periodic table and their electronic configurations structures Special emphasis has been put on the discussion of the correlation between the structure and properties of elements compound The book caters to the requirements of Bachelor in Science Pass courses With detailed discussion on several advanced topics the

students of Bachelor in Science Honours and Masters in Science would also find it extremely useful

Introduction to Modern Chemistry: Student Manual Eugene Meyer, 1979 *Redox Chemistry and Interfacial Behavior of Biological Molecules* Glenn Dryhurst, K Niki, 2012-12-06 The papers in this book were presented at the Third International Symposium on Redox Mechanisms and Interfacial Properties of Molecules of Biological Importance held in Honolulu Hawaii between October 19 23 1987 This Symposium was held as part of the 172nd Meeting of The Electrochemical Society which was cosponsored by The Electrochemical Society of Japan with the cooperation of The Japan Society of Applied Physics The aim of the Symposium was to bring together a group of electrochemists and bio medical scientists with interests in electrochemistry from around the world to present their most current research results and or to present up to date reviews of current areas of research activity It is quite clear from the diversity of topics covered in the various papers that electrochemistry and electrochemical techniques and principles have much to contribute to our understanding of many important biochemical phenomena For example electrochemical studies are providing important insights into the redox properties of biomolecules ranging from relatively small organic molecules such as indoleamine neurotransmitters to very large organic organometallic molecules which include various redox enzymes or model enzyme systems Many of the most powerful analytical techniques are now being coupled to electrodes to monitor potential controlled behaviors of biological molecules at charged interfaces Electrochemical techniques are now being developed which permit extraordinarily small electrodes to be inserted into single cells to monitor electroactive biomolecules Other microelectrodes are being employed to control cell growth and to manipulate single cells

Fundamentals of Chemistry: A Modern Introduction Frank Brescia, 2012-12-02 Fundamentals of Chemistry Third Edition introduces the reader to the fundamentals of chemistry including the properties of gases atomic and molecular weights and the first and second laws of thermodynamics Chemical equations and chemical arithmetic are also discussed along with the structure of atoms chemical periodicity types of chemical bonds and condensed states of matter This book is comprised of 26 chapters and begins with a historical overview of chemistry and some terms which are part of the language of chemists Separation and purification are covered in the first chapter while the following chapters focus on atomic and molecular weights stoichiometry the structure of atoms and types of chemical bonds The molecular orbital MO theory of bonding galvanic cells and chemical thermodynamics are considered next Separate chapters are devoted to MO theory of covalent and metallic bonding orbital hybridization intermolecular forces acids and bases ionic equilibrium calculations and polymers and biochemicals This monograph is intended for chemistry students

Principles of Modern Chemistry David W. Oxtoby, Norman H. Nachtrieb, 1996 PRINCIPLES OF MODERN CHEMISTRY has dominated the honors and high mainstream general chemistry courses and is considered the standard for the course The fifth edition is a substantial revision that maintains the rigor of previous editions but reflects the exciting modern developments taking place in chemistry today Authors David W Oxtoby and H P Gillis provide a unique approach to learning chemical principles that

emphasizes the total scientific process from observation to application placing general chemistry into a complete perspective for serious minded science and engineering students Chemical principles are illustrated by the use of modern materials comparable to equipment found in the scientific industry Students are therefore exposed to chemistry and its applications beyond the classroom This text is perfect for those instructors who are looking for a more advanced general chemistry textbook

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Longman Science Chemistry 10 Kohli Nitin, 2008-09 Recent Advances in Redox Active Plant and Microbial Products Claus Jacob, Gilbert Kirsch, Alan Slusarenko, Paul G. Winyard, Torsten Burkholz, 2014-11-25 Nature endows us with a treasure chest of Green Gold full of amazing redox active substances which interfere with numerous biological processes in our own body in animals bacteria fungi and plants Whilst such natural products are all around and also in us we still do not fully understand how these compounds actually work This book attempts to resolve some of the mysteries and riddles associated with such products Written by more than thirty international experts from academia and industry it places a focus on modern developments in this field and considers such natural products from various angles from their isolation and characterization all along to product development and commercialization Throughout the reader will be confronted with modern approaches which enable the efficient identification and isolation of new natural products help to elucidate their mode s of action and permit practical uses in Medicine Cosmetics Agriculture Industry and as functional foods

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Chemistry in Daily Life Dr. Benvikram Barman, Dr. Bhupendra Singh Banjare, Dr. Dolly Baghel, Dr. Dinesh Kumar Tandan, Mr. Naveen Verma, Mr. Rupendra Kumar Sahu,2025-08-12 Chemistry in Daily Life is a comprehensive textbook designed to connect fundamental chemical principles with everyday applications In this book have 23 chapter each chapter explain with suitable figure and in last section of each chapter have multiple type question with answer This book explores how chemistry influences food environment health and industry It covers the composition and quality analysis of dairy products beverages food additives including artificial sweeteners Flavors preservatives colorants and contaminants Chapters paints dyes pigments and their industrial applications Air and water pollution are observed finished topics like air pollutants greenhouse gases acid rain and wastewater treatment Readers gain understanding into water purification methods the hydrologic cycle and soil chemistry The role of fertilizers and their classifications is explained alongside the chemistry of essential biomolecules like vitamins carbohydrates oils Medicinal chemistry is detailed discussions on analgesics antibiotics antimalarials other drugs The book also introduces concepts in chemical kinetics thermodynamics and energy systems including fuel cells batteries solar energy Finally environmental challenges such as plastic waste heavy metal pollution are addressed with eco friendly solutions Rich in real life examples diagrams exercises this book is perfect for students educators and anyone interested how chemistry shapes our world *NEET UG Chemistry Study Notes with Theory + Practice MCQs for Complete Preparation | Based on New Syllabus as per NMC* EduGorilla Prep Experts, *National 5 Chemistry* Barry McBride,Fran Macdonald,John Anderson,2015-10-30 A full course textbook for the new National 5 Chemistry syllabus endorsed by SQA This book is designed to act as a valuable resource for pupils studying National 5 Chemistry It provides a core text which adheres closely to the SQA syllabus with each section of the book matching a unit of the syllabus and each chapter corresponding to a content area It is an ideal and comprehensive teaching and learning resource for National 5 Chemistry In addition to the core text the book contains a variety of special features For Interest Key Terms Activities Worked Examples Questions Prescribed Practical Activities Summary and Checklist for Revision The only textbook for the National 5 Chemistry syllabus offered by SQA as examined 2014 onwards Bestselling author team with extremely high reputation for Scottish Chemistry titles Full colour presentation and motivating text design to encourage student enthusiasm

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