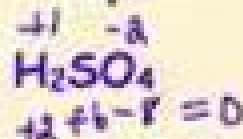


Examples



$$\text{H} = (+1)2$$

$$\text{S} = \text{no rule directly applied}$$

$$\text{O} = (-2)4$$

$$\underline{+2}$$

$$\underline{+6}$$

$$\underline{-8}$$

- So therefore, S must be +6 in order to make the whole compound equal zero!



$$\text{K} = (+1)2$$

$$\text{Cr} = (\text{no rule directly applied})2$$

$$\text{O} = (-2)7$$

$$\underline{+2}$$

$$\underline{(+6)2}$$

$$\underline{-14}$$

- Due to the fact that Cr does not have a rule directly assigning it an oxidation number, a number is assigned to it to make the formula electrically neutral. The only concern is that we take into account that it is Cr has a subscript of 2

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**Mahendra Rai, Shagufta Khan, Aarti
Belgamwar**



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