

# THE "OFFICIAL" CHEMISTRY 12 REDOX & ELECTROCHEMISTRY STUDY GUIDE

1. **DO ALL THE QUESTIONS** in this booklet. These are actual Provincial Exam questions! Your own provincial exam and unit test will include questions similar to the ones in this booklet!
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5. **Check Off the STATUS box on the PRESCRIBED LEARNING OUTCOMES sheet.** I have tried to organize the questions in the identical sequence to which they appear on your Redox/Electrochemistry Prescribed Learning Outcome sheet. By doing this, you can be confident that you know everything you need to know for both the UNIT EXAM and PROVINCIAL EXAM!

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## INTRODUCTION

1. **501** Which one of the following is the strongest reducing agent?

☒ A.  $\text{I}^-$       B.  $\text{F}^-$       C.  $\text{Cl}^-$       D.  $\text{Br}^-$

2. **501** In the reaction below:



A.  $\text{I}_2$       B.  $\text{I}^-$       C.  $\text{H}^+$       ☒ D.  $\text{ClO}_3^-$

3. **501** Which of the following is the **strongest** reducing agent?

A. cobalt  
B. copper  
☒ C. calcium  
D. chromium

4. **501** An oxidizing agent is

A. reduced as it loses electrons.  
☒ B. reduced as it gains electrons.  
C. oxidized as it loses electrons.  
D. oxidized as it gains electrons.

5. **501** Which of the following is the **strongest** oxidizing agent?

☒ A. Mn  
B.  $\text{Mn}^{2+}$   
☒ C. acidified  $\text{MnO}_2$   
☒ D. acidified  $\text{MnO}_4^-$

# Redox Reaction Chem Study Guide Answers

**Shasha Hu**



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