

BALANCING REDOX REACTIONS

STEP BY STEP IN TWO WAYS...



BOTH START THE SAME WAY

1. ASSIGN OXIDATION NUMBERS TO EACH ATOM



NOTE: OXIDATION NUMBERS SUM UP TO ZERO ALWAYS

ALWAYS HAVE AN OX TO GO AND REDUCED STATE FIRST

2. BALANCE EACH HALF REACTION SEPARATELY

3. SEPARATE YOUR EQUATION AS TWO HALF REACTIONS (BALANCING EACH ATOM SEPARATELY IN THIS STEP)

NOTE: ALWAYS OX. 1st & REDUCED STATE SECOND, & OX. SECOND



HALF-REACTION METHOD (in acidic conditions)

4. BALANCE OXIDATION



5. BALANCE REDUCTION



6. BALANCE CHARGES WITH ELECTRONS

NOTE: ALWAYS MAKE POSITIVE FIRST AND ALWAYS MAKE POSITIVE LAST



7. MAKE SURE THE NUMBER OF ELECTRONS IS THE SAME

NOTE: ALWAYS MAKE POSITIVE LAST AND ALWAYS MAKE POSITIVE LAST



8. PUT BOTH REACTIONS BACK TOGETHER



9. CHECK FOR ANYTHING THAT IS BOTH A PRODUCT AND REACTANT

NOTE: ALWAYS BALANCE FIRST AND ALWAYS MAKE POSITIVE LAST



OXIDATION NUMBER METHOD (in acidic conditions)

4. BALANCE OXIDATION SEPARATELY WITH OTHER FIRST

NOTE: ALWAYS MAKE POSITIVE FIRST AND ALWAYS MAKE POSITIVE LAST



7. MAKE SURE THE NUMBER OF ELECTRONS IS THE SAME

NOTE: ALWAYS MAKE POSITIVE FIRST AND ALWAYS MAKE POSITIVE LAST



8. PUT BOTH REACTIONS BACK TOGETHER AND CHECK FOR ELECTRONS



9. BALANCE OXIDATION



10. BALANCE REDUCTION



BUT WHAT ABOUT BASIC CONDITIONS?

WHEN WE SAY "ACIDIC" OR "BASIC" WE ARE USING TWO APPROXIMATE DEFINITIONS:

ACIDIC CONDITIONS HAVE H⁺ AND BASIC CONDITIONS HAVE OH⁻

SO WHEN WE FIND OUR REACTION FROM ABOVE WITH A REACTION PERFORMED IN BASIC CONDITIONS:

8. ADD THE NUMBER OF OH⁻ TO THE NUMBER OF H⁺ IN EACH SIDE

NOTE: ALWAYS MAKE POSITIVE FIRST AND ALWAYS MAKE POSITIVE LAST



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H. Stephen Stoker, Danny V. White



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Reviewing **Redox Reactions Teacher Guide And Answers**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Redox Reactions Teacher Guide And Answers**," an enthralling opus penned by a very acclaimed wordsmith, readers set about an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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