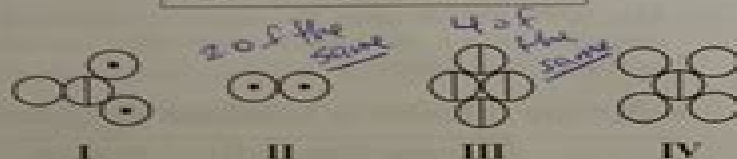
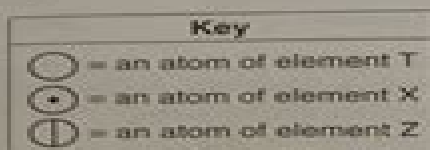


39 Given four particle models:



Which two models can be classified as elements?

- (1) I and II  
 (2) I and IV  
 (3) II and III  
 (4) II and IV

40 After being thoroughly stirred at  $10^{\circ}\text{C}$ , which mixture is heterogeneous?

- (1) 25.0 g of KCl and 100. g of  $\text{H}_2\text{O}$  *unsat.*  
 (2) 25.0 g of  $\text{KNO}_3$  and 100. g of  $\text{H}_2\text{O}$  *supersat.*  
 (3) 25.0 g of NaCl and 100. g of  $\text{H}_2\text{O}$  *unsat.*  
 (4) 25.0 g of  $\text{NaNO}_3$  and 100. g of  $\text{H}_2\text{O}$  *unsat.*

41 Which two compounds are electrolytes?

- (1) KOH and  $\text{CH}_3\text{COOH}$   
 (2) KOH and  $\text{C}_3\text{H}_{12}$   
 (3)  $\text{CH}_3\text{OH}$  and  $\text{CH}_3\text{COOH}$   
 (4)  $\text{CH}_3\text{OH}$  and  $\text{C}_3\text{H}_{12}$

42 Which statement explains why a  $\text{CO}_2$  molecule is nonpolar?

- (1) Carbon and oxygen are both nonmetals.  
 (2) Carbon and oxygen have different electronegativities.  
 (3) The molecule has a symmetrical distribution of charge.  
 (4) The molecule has an asymmetrical distribution of charge.

43 Which temperature change indicates an increase in the average kinetic energy of the molecules in a sample?

- (1)  $15^{\circ}\text{C}$  to  $298\text{ K}$   $\uparrow$   
 (2)  $37^{\circ}\text{C}$  to  $273\text{ K}$   $\downarrow$   
 (3)  $305\text{ K}$  to  $0^{\circ}\text{C}$   $\downarrow$   
 (4)  $355\text{ K}$  to  $25^{\circ}\text{C}$   $\downarrow$

44 Given the particle diagram:



Which substance at STP can be represented by this particle diagram? *Solid*

- (1)  $\text{N}_2$  (g)  
 (2)  $\text{H}_2$  (g)  
 (3)  $\text{Mg}$  (s)  
 (4)  $\text{Kr}$  (g)

45 Which type of equilibrium exists in a sealed flask containing  $\text{Br}_2(\ell)$  and  $\text{Br}_2(\text{g})$  at  $298\text{ K}$  and  $1.0\text{ atm}$ ?

- (1) static phase equilibrium  
 (2) static solution equilibrium  
 (3) dynamic phase equilibrium  
 (4) dynamic solution equilibrium

46 What are the products when potassium hydroxide reacts with hydrochloric acid?

- (1)  $\text{KH}(\text{s})$ ,  $\text{Cl}^-(\text{aq})$ , and  $\text{OH}^-(\text{aq})$   
 (2)  $\text{K}(\text{s})$ ,  $\text{Cl}_2(\text{g})$ , and  $\text{H}_2\text{O}(\ell)$   
 (3)  $\text{KCl}(\text{aq})$  and  $\text{H}_2\text{O}(\ell)$   
 (4)  $\text{KOH}(\text{aq})$  and  $\text{Cl}_2(\text{g})$

47 In a titration, 20.0 milliliters of a  $0.150\text{ M}$   $\text{NaOH}(\text{aq})$  solution exactly neutralizes 24.0 milliliters of an  $\text{HCl}(\text{aq})$  solution. What is the concentration of the  $\text{HCl}(\text{aq})$  solution?

- (1)  $0.125\text{ M}$   
 (2)  $0.180\text{ M}$   
 (3)  $0.250\text{ M}$   
 (4)  $0.360\text{ M}$

48 What fraction of a  $\text{Sr-90}$  sample remains unchanged after 87.3 years?

- (1)  $\frac{1}{2}$   
 (2)  $\frac{1}{3}$   
 (3)  $\frac{1}{4}$   
 (4)  $\frac{1}{8}$

*Handwritten calculation:*  

$$3 = \frac{87.3}{29.14}$$

$$\frac{1}{2} = \frac{1}{8}$$

# Regents Chemistry Pearson Review Answers

**Halina Chodkiewicz, Piotr  
Steinbrich, Małgorzata Krzemińska-  
Adamek**

## **Regents Chemistry Pearson Review Answers:**

**Chemistry and Industry Review** ,1958      **Chemistry** Patrick Kavanah,2004-08-30      Working with Text and Around Text in Foreign Language Environments Halina Chodkiewicz,Piotr Steinbrich,Małgorzata Krzemińska-Adamek,2016-06-28  
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## **Table of Contents Regents Chemistry Pearson Review Answers**

1. Understanding the eBook Regents Chemistry Pearson Review Answers
  - The Rise of Digital Reading Regents Chemistry Pearson Review Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Regents Chemistry Pearson Review Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Regents Chemistry Pearson Review Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Regents Chemistry Pearson Review Answers
  - Personalized Recommendations
  - Regents Chemistry Pearson Review Answers User Reviews and Ratings
  - Regents Chemistry Pearson Review Answers and Bestseller Lists
5. Accessing Regents Chemistry Pearson Review Answers Free and Paid eBooks
  - Regents Chemistry Pearson Review Answers Public Domain eBooks
  - Regents Chemistry Pearson Review Answers eBook Subscription Services
  - Regents Chemistry Pearson Review Answers Budget-Friendly Options
6. Navigating Regents Chemistry Pearson Review Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Regents Chemistry Pearson Review Answers Compatibility with Devices
  - Regents Chemistry Pearson Review Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Regents Chemistry Pearson Review Answers
  - Highlighting and Note-Taking Regents Chemistry Pearson Review Answers
  - Interactive Elements Regents Chemistry Pearson Review Answers
8. Staying Engaged with Regents Chemistry Pearson Review Answers

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Regents Chemistry Pearson Review Answers
- 9. Balancing eBooks and Physical Books Regents Chemistry Pearson Review Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Regents Chemistry Pearson Review Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Regents Chemistry Pearson Review Answers
  - Setting Reading Goals Regents Chemistry Pearson Review Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Regents Chemistry Pearson Review Answers
  - Fact-Checking eBook Content of Regents Chemistry Pearson Review Answers
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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