

2. Figure 1 show a part of the periodic table.

	1	2													13	14	15	16	17	18
1																				
2																P		Q		
3		R																	S	T
4																				
5																				
6																				

Figure 1

Based on the Figure 1, answer the following questions.

- (a) What is the element represented by P? (1 marks)
- (b) On the figure above, mark X in the box for an alkali metal (1 marks)
- (c) State one characteristic of element U. (1 marks)
- (d) Write the electron arrangement for S. (1 marks)
- (e) (i) Which element is not chemically reactive? (1 marks)
- (ii) State one application of the element in (e)(i) (1 marks)
- (f) (i) Write the chemical equation for the reaction between R and Q (1 marks)
- (ii) State briefly the transfer of electron which occur during the reaction between R and Q. (2 marks)

JAWAPAN

1 (a)	Carbon	1
(b)	(any box in column for group 1)	1
(c)	Form coloured compounds	1
(d)	2.8.7	1
(e) (i)	T	1
(ii)	Fill electric bulb	1
(f) (i)	$2R + Q_2 \rightarrow 2RQ$	1
(ii)	- R release two electrons to Q - Q receive two electron from R	1 1
		9

Regents Periodic Table Questions

Yicheng Fang



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Table of Contents Regents Periodic Table Questions

1. Understanding the eBook Regents Periodic Table Questions
 - The Rise of Digital Reading Regents Periodic Table Questions
 - Advantages of eBooks Over Traditional Books
2. Identifying Regents Periodic Table Questions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Regents Periodic Table Questions
 - User-Friendly Interface
4. Exploring eBook Recommendations from Regents Periodic Table Questions
 - Personalized Recommendations
 - Regents Periodic Table Questions User Reviews and Ratings
 - Regents Periodic Table Questions and Bestseller Lists
5. Accessing Regents Periodic Table Questions Free and Paid eBooks
 - Regents Periodic Table Questions Public Domain eBooks
 - Regents Periodic Table Questions eBook Subscription Services
 - Regents Periodic Table Questions Budget-Friendly Options
6. Navigating Regents Periodic Table Questions eBook Formats

- ePub, PDF, MOBI, and More
- Regents Periodic Table Questions Compatibility with Devices
- Regents Periodic Table Questions Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Regents Periodic Table Questions
 - Highlighting and Note-Taking Regents Periodic Table Questions
 - Interactive Elements Regents Periodic Table Questions
- 8. Staying Engaged with Regents Periodic Table Questions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Regents Periodic Table Questions
- 9. Balancing eBooks and Physical Books Regents Periodic Table Questions
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Regents Periodic Table Questions
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Regents Periodic Table Questions
 - Setting Reading Goals Regents Periodic Table Questions
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Regents Periodic Table Questions
 - Fact-Checking eBook Content of Regents Periodic Table Questions
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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