Figure 1 show a part of the periodic table.

T	12										13	14	1.5	16	17	7.0
												P	T	Q		1
	R														S	T
	-	_	-	+	-		-		-	-	_	-	-	$\vdash$	-	-
				-	-	-		-		-			4	1		+

Figure 1

Based on the Figure 1, answer the following questions.

- (a) What is the element represented by P?
- ......(1 marks)
- (b) On the figure above, mark X in the box for an alkali metal (1 marks)
- (c) State one characteristic of element U.
- (d) Write the electron arrangement for S.
- ......(1 marks)
- (e) (i) Which element is not chemically reactive?
  - (ii)State one application of the element in (e)(i)
    (1 marks)
- (f) (i) Write the chemical equation for the reaction between R and Q
  - (1 marks)
  - (ii)State briefly the transfer of electron which occur during the reaction between R and Q.

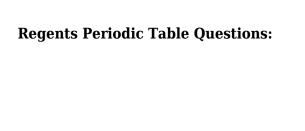
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#### **JAWAPAN**

L(a)	Carbon	1
(b)	(any box in column for group 1)	1
(c)	Form coloured compounds	1
(d)	2.8.7	1
(e) (i) (ii)	T Fill electric bulb	1
(f) (i) (ii)	2R + Q <sub>1</sub> → 2RQ  R release two electrons to Q  Q receive two electron from R	1 1 1
		9

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