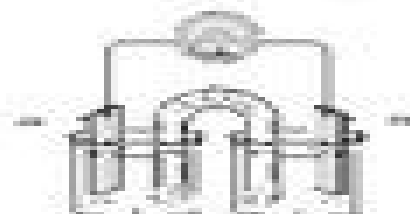


ANS: A

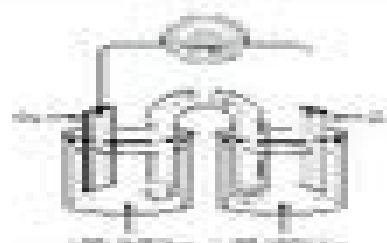
1. As the cell operates, the electrons flow towards the



- A) Zn electrode and the cell voltage increases over time
- B) Pb electrode and the cell voltage decreases over time
- C) Zn electrode and the cell voltage decreases over time
- D) Pb electrode and the cell voltage remains constant over time

ANS: B

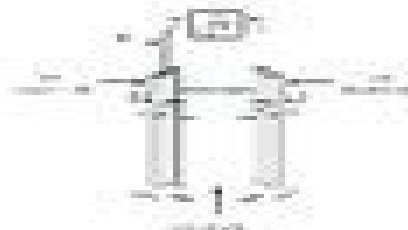
1. What happens to the zinc electrode?



- A) Mass increases as it is reduced
- B) Mass decreases as it is reduced
- C) Mass increases as it is oxidized
- D) Mass decreases as it is oxidized

ANS: D

1. What reaction occurs at the cathode?



Regents Chemistry Electrochemical Cell Answers

G Thomas



Regents Chemistry Electrochemical Cell Answers:

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