

## Unit 3: Acids and Bases

This unit activity will help you meet these chemistry goals:

- 1. **Science Inquiry.**—This unit explores the nature of acids, bases, acids and bases reacting, including their properties and structures, and collecting information that can be used to compare acids and bases.
- 2. **Chemical Skills.**—This unit employs techniques for measuring and analyzing.

### Introduction

Acids and bases are common to many areas of biological and environmental systems. They are also important in many other areas of science, including chemistry. The properties of acids and bases are determined by their structure and composition, and are influenced by the environment in which they are found. Understanding the properties of acids and bases is important for many other areas of science, including chemistry, biology, and environmental science.

- Is the compound an acid or a base?
- Is the compound strong or weak?

In this activity, you will explore the nature of acids and bases, and how they react. You will also explore how acids and bases are used in many other areas of science, including chemistry, biology, and environmental science. You will also explore how acids and bases are used in many other areas of science, including chemistry, biology, and environmental science.

### Objectives and Analysis

**Task 1: Acids and Bases in real world applications**

Many research teams have explored the nature of acids and bases, and how they react. This unit activity will help you explore the nature of acids and bases, and how they react. You will also explore how acids and bases are used in many other areas of science, including chemistry, biology, and environmental science.

- 1. For the purpose of this activity, acids and bases are defined as follows: acids are substances that donate protons (H<sup>+</sup>), and bases are substances that accept protons (H<sup>+</sup>).

**Key point: Acids and Bases**

Acids and bases are defined as follows: acids are substances that donate protons (H<sup>+</sup>), and bases are substances that accept protons (H<sup>+</sup>). Acids and bases are also defined as follows: acids are substances that donate protons (H<sup>+</sup>), and bases are substances that accept protons (H<sup>+</sup>).

# Plato Chemistry B Unit 5

**Terry C. Jones**



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