

# Aqueous Reactions: Neutralization

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1. What chemical species are present in a sample of nitric acid? Make sure to include all atoms, molecules, or ions present.

Hydrogen ions/Hydronium ions  $\text{H}^+/\text{H}_3\text{O}^+$

Nitrate ions  $\text{NO}_3^-$

Water molecules  $\text{H}_2\text{O}$

2. What chemical species are present in a sodium hydroxide solution? Make sure to include all atoms, molecules, or ions present.

Sodium ions  $\text{Na}^+$

Hydroxide ions  $\text{OH}^-$

Water molecules  $\text{H}_2\text{O}$

3. In the simulation, you mix nitric acid with sodium hydroxide. At the instant that these two solutions are mixed (before there is time for a reaction to occur), list all species present.

Hydrogen ions/Hydronium ions  $\text{H}^+/\text{H}_3\text{O}^+$

Nitrate ions  $\text{NO}_3^-$

Sodium ions  $\text{Na}^+$

Hydroxide ions  $\text{OH}^-$

Water molecules  $\text{H}_2\text{O}$

4. Draw a picture to represent the system at the instant that the two solutions are mixed.

# Neutralization Reactions Answers Key

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