

Write the number of each kind of atom in the following compounds. (2 points each)

33.) MgSO_4 Mg S O

- a. 4-4-4 b. 1-4-4
c. 1-1-4 d. 1-1-1

34.) $\text{Cu}_2(\text{PO}_3)_4$ Cu P O

- a. 2-2-4 b. 2-2-6
c. 2-1-6 d. 2-2-6

35.) $(\text{NH}_4)_2\text{SO}_4$ N S O

- a. 2-4-2-7 b. 1-4-2-7
c. 4-4-2-7 d. 4-4-2-7

Draw a cooling curve phase diagram.

36.) H_2O

- a. Solid b. Gas
c. Liquid d. Melting
e. Freezing

37.) CO_2

- a. Solid b. Gas
c. Liquid d. Melting
e. Freezing

38.) CO_2

- a. Solid b. Gas
c. Liquid d. Melting
e. Freezing

39.) Which section has the highest energy?

- a. A to B b. B to C
c. C to D d. D to E
e. E to F

40.) Which section has the lowest energy?

- a. A to B b. B to C
c. C to D d. D to E
e. E to F



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