

Physical Science EOCT Study Guide – Domain 3

Physical Energy, Force, and Motion

Standards: PS.3.1 and 3.2

1. Students will analyze transformations and flow of energy within a system.
- Define thermal energy, heat, and temperature. Analyze the relationship between the amount of thermal energy and the temperature for two objects.
 - Define heat. Theoretical heating curve showing phase transitions for water. Calculate the heat required to change the state of water from solid to liquid, liquid to gas, and solid to gas.
 - Define specific heat. The amount of heat needed to raise the temperature of 1 kg of a substance by 1°C.
 - What is the formula for determining that change in state that energy can take?
 $Q = mc\Delta T$ or $Q = mL$ or $Q = m\Delta T$ $\Delta T = T_2 - T_1$
 - A sample of 0.5 kg of water at 20°C and specific heat of 4186 J/kg°C. Calculate the heat required to raise the temperature of the water to 80°C.
 $Q = mc\Delta T$ $Q = (0.5 \text{ kg})(4186 \text{ J/kg°C})(80°C - 20°C) = 104,650 \text{ J}$
 - Explain conduction, convection, and radiation. Give examples of each.
 Conduction: heat transfer by direct contact. Example: a metal spoon in a pot of soup.
 Convection: heat transfer by movement of a fluid. Example: a pot of boiling water.
 Radiation: heat transfer by electromagnetic waves. Example: the sun's rays.

2. Explain the following phase change diagrams.



3. Students will determine relationships among force, mass and motion.
- Explain the difference between distance and displacement.
 Distance is the total path length traveled.
 Displacement is the straight-line distance from start to end.
 - What is the difference between average speed and instantaneous speed?
 Average speed = $\frac{\text{Total Distance}}{\text{Total Time}}$
 Instantaneous speed = $\frac{d}{dt}$
 - What is the difference between velocity and speed?
 Velocity is speed with direction.
 Speed is the magnitude of velocity.

$$v = \frac{d}{t}$$

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M Mark



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