

Nuclear Fission and Fusion

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As you read this section, keep these questions in mind:

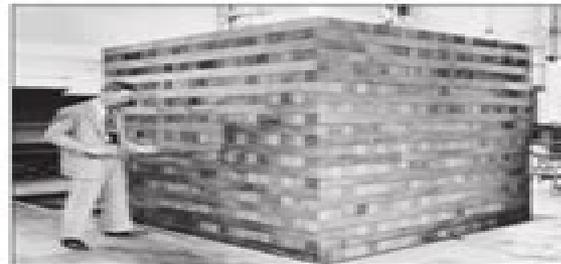
- What holds the nucleus of an atom together?
- What happens when the nucleus of a heavy atom splits apart?
- What happens when the nucleus of a small atom joins with the nucleus of another small atom?

Who Discovered Nuclear Fission?

In 1939, German scientists Otto Hahn and Fritz Strassman were trying to make uranium atoms with heavier nuclei. To do this, they used a device that shot neutrons at a sample of uranium. They hoped that at least a few uranium nuclei would gain one or more neutrons. To their surprise, new elements formed. Instead of uranium isotopes, they detected barium and krypton. They thought they had made a mistake.

Soon after, a scientist named Lise Meitner and her nephew Otto Frisch read the results of the experiments. Meitner realized that Hahn and Strassman had not made a mistake. They had actually split the uranium nuclei into smaller elements. ☑

This nuclear reactor was used in the late 1940s and early 1950s to study controlled nuclear fission.



In the early 1940s, Enrico Fermi and other scientists at the University of Chicago built a stack of graphite and uranium blocks. These materials formed the nuclear reactor that was used to create the first controlled nuclear fission chain reaction. This work launched the Manhattan Project, which led to the creation of nuclear weapons.

READING TOOLBOX

Define As you read, write down any terms in the section that you do not understand. Use the section vocabulary boxes or a dictionary to find the definitions of these terms. Write the definitions in your own words in the text margin.

READING CHECK

1. Explain What did Lise Meitner conclude about the work of Hahn and Strassman?

Physical Science Fission And Fusion Worksheet

S Ashworth



Physical Science Fission And Fusion Worksheet:

Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction *Energy* Judith H. Higgins, 1979 *Resources in education*, 1987-07 **Films and Other Materials for Projection** Library of Congress, 1978 **National Union Catalog**, 1973 **Earth Science**, 2001 **Resources in Education**, 1987 *Focus on Earth Science*, 2001 Nuclear Fusion and Fission Fiona Young-Brown, 2016-07-15 Nuclear Fusion and Fission delves into nuclear physics and the scientists responsible for the discovery of splitting and fusing an atom The book begins with the very basic building blocks of science breaking down the different types of energy and how we use them the materials that make up an atom and our search for the perfect renewable energy source Set against the cultural backdrop of World War II later chapters follow each significant theory that led to the creation of the world's most dangerous weapon as well as some of its most widely used medical and food production processes today **What's the Difference Between Fission and Fusion? | Children's Physics of Energy** Baby Professor, 2017-02-15 Although fission and fusion are both nuclear reactions they are miles apart in terms of definitions Fission splits while fusion combines You can use this child friendly book of physics to teach the principles of nuclear reactions in the simplest manner The carefully selected images will serve as additional guide to influence the visual effectiveness of this resource Get a copy today Bulletin of the Atomic Scientists, 1970-06 The Bulletin of the Atomic Scientists is the premier public resource on scientific and technological developments that impact global security Founded by Manhattan Project Scientists the Bulletin's iconic Doomsday Clock stimulates solutions for a safer world English for Specific Purposes Ronald Mackay, 1978 **Government Reports Announcements & Index**, 1979 Fission, Fusion and the Energy Crisis Stanley Ernest Hunt, 1974 Principles of Fusion Energy A. A. Harms, Klaus F. Schoepf, David Ross Kingdon, 2000 This textbook accommodates the two divergent developmental paths which have become solidly established in the field of fusion energy the process of sequential tokamak development toward a prototype and the need for a more fundamental and integrative research approach before costly design choices are made Emphasis is placed on the development of physically coherent and mathematically clear characterizations of the scientific and technological foundations of fusion energy which are specifically suitable for a first course on the subject Of interest therefore are selected aspects of nuclear physics electromagnetics plasma physics reaction dynamics materials science and engineering systems all brought together to form an integrated perspective on nuclear fusion and its practical utilization The book identifies several distinct themes The first is concerned with preliminary and introductory topics which relate to the basic and relevant physical

processes associated with nuclear fusion Then the authors undertake an analysis of magnetically confined inertially confined and low temperature fusion energy concepts Subsequently they introduce the important blanket domains surrounding the fusion core and discuss synergetic fusion fission systems Finally they consider selected conceptual and technological subjects germane to the continuing development of fusion energy systems

Cold Nuclear Fusion Florian Ion Petrescu,2012-07-01 Nuclear fusion is the process by which two or more atomic nuclei join together or fuse to form a single heavier nucleus During this process matter is not conserved because some of the mass of the fusing nuclei is converted to energy which is released The binding energy of the resulting nucleus is greater than the binding energy of each of the nuclei that fused to produce it Fusion is the process that powers active stars Creating the required conditions for fusion on Earth is very difficult to the point that it has not been accomplished at any scale for protium the common light isotope of hydrogen that undergoes natural fusion in stars In nuclear weapons some of the energy released by an atomic bomb fission bomb is used for compressing and heating a fusion fuel containing heavier isotopes of hydrogen and also sometimes lithium to the point of ignition At this point the energy released in the fusion reactions is enough to briefly maintain the reaction Fusion based nuclear power experiments attempt to create similar conditions using far lesser means although to date these experiments have failed to maintain conditions needed for ignition long enough for fusion to be a viable commercial power source There are many experiments examining the possibility of fusion power for electrical generation Nuclear fusion has great potential as a sustainable energy source This is due to the abundance of hydrogen on the planet and the inert nature of helium the nucleus which would result from the nuclear fusion of hydrogen atoms Unfortunately a controlled nuclear fusion reaction has not yet been achieved due to the temperatures required to sustain one In hot fusion it need a temperature of 4000 million degrees Without a minimum of 3000 million degrees we can t make the hot fusion reaction to obtain the nuclear power Today we have just 150 million degrees made To replace the lack of necessary temperature it uses various tricks Because obtaining the necessary huge temperature for hot fusion is still difficult it is time to focus us on cold nuclear fusion We need to bomb the fuel with accelerated deuterium nuclei The fuel will be made from heavy water and lithium The optimal proportion of lithium will be tested It would be preferable to keep fuel in the plasma state Research into developing controlled thermonuclear fusion for civil purposes also began in earnest in the 1950s and it continues to this day Two projects the National Ignition Facility and ITER are in the process of reaching breakeven after 60 years of design improvements developed from previous experiments The best results were obtained with the Tokamak type installations

Nuclear Fission and Fusion in a Random-walk Model Martin Albertsson,2021 **FISSION, FUSION AND SPIES** ROSCOE B. WHITE,2025

Nuclear Physics for Morons in Slides R. E. D. Dot PUBLICATIONS,2017-08-15 Basic understanding of nuclear physics in slides Very instructive and easy to understand for the beginner **The Curve of Binding Energy** Paul F. Kisak,2016-05-31

Nuclear binding energy is the energy that would be required to disassemble the nucleus of an atom into its component parts

These component parts are neutrons and protons which are collectively called nucleons. The binding energy of nuclei is due to the attractive forces that hold these nucleons together and this is usually a positive number since most nuclei would require the expenditure of energy to separate them into individual protons and neutrons. The mass of an atomic nucleus is usually less than the sum of the individual masses of the constituent protons and neutrons according to Einstein's equation $E = mc^2$ and this missing mass is known as the mass defect and represents the energy that was released when the nucleus was formed. The term nuclear binding energy may also refer to the energy balance in processes in which the nucleus splits into fragments composed of more than one nucleon. If new binding energy is available when light nuclei fuse or when heavy nuclei split either process can result in release of this binding energy. This energy may be made available as nuclear energy and can be used to produce electricity as in nuclear power or in a nuclear weapon. When a large nucleus splits into pieces excess energy is emitted as photons (gamma rays) and as the kinetic energy of a number of different ejected particles (nuclear fission products). The nuclear binding energies and forces are on the order of a million times greater than the electron binding energies of light atoms like hydrogen. The mass defect of a nucleus represents the mass of the energy of binding of the nucleus and is the difference between the mass of a nucleus and the sum of the masses of the nucleons of which it is composed.

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