# Physical Science Chapter 6 Study Guide

- Atomic Theory of Matter
  - Every element consists of tiny particles called atoms
  - All atoms of a particular element have the same properties
  - Different elements have different properties because their atoms are different
  - Atoms of different elements can combine in specific ways to form compounds
  - Chemical processes are the result of the rearrangement, combination, or separation of atoms
- Abbreviation given to each element -chemical symbol—Know the bold print chemical symbols
- Parts of an atom
  - Nucleus—dense central core composed of protons and neutrons
  - Electrons surround the nucleus
  - Protons and neutrons are made up of small particles called quarks
  - Protons have positive charge and electrons have a negative charge
- Isotopes—atoms that are of the same element but have different number of neutrons
- Electron shells—regions around the nucleus that represent the energy level of electrons
- lons—atoms with a charge
  - Cation—atom with positive charge
  - Anion—atom with negative charge

# **Physical Science Chapter 6 Vocabulary**

Jessica J Manson

**Physical Science Chapter 6 Vocabulary:** 

Unveiling the Power of Verbal Artistry: An Psychological Sojourn through Physical Science Chapter 6 Vocabulary

In a global inundated with displays and the cacophony of quick communication, the profound power and psychological resonance of verbal artistry usually disappear in to obscurity, eclipsed by the continuous assault of sound and distractions. Yet, nestled within the lyrical pages of **Physical Science Chapter 6 Vocabulary**, a captivating work of fictional splendor that pulses with fresh feelings, lies an unforgettable trip waiting to be embarked upon. Written by way of a virtuoso wordsmith, that enchanting opus courses visitors on an emotional odyssey, delicately revealing the latent potential and profound affect embedded within the complex web of language. Within the heart-wrenching expanse of this evocative analysis, we can embark upon an introspective exploration of the book is central themes, dissect their charming writing style, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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