

**First Chapter Short Questions (Fsc Part I)**  
**Malik Xufyan**  
**Msc Chemistry**

**Q4:** What are ions? Under What condition are they produced?

**Ans:**

**Ions**

Ions are produced by gaining or loosing electrons .

ions are those substance which have positive or negative ion

cation are those specie which have positive charge by removal of electron

Anion are those species which have negative charges by gaining of electrons

**Conditions:**

Ions are produced by

- (i) By dissolving ionic compounds in water
- (ii) By X-rays
- (iii) In mass spectrometry
- (iv) By removing or adding electron in atom

**Q8:** Justify the following statements:

**(a) 23 g of sodium and 238g of uranium have equal number of atoms in them**

**Ans :**

23g of Na = 1 mole of Na =  $6.02 \times 10^{23}$  atoms of Na

238g of U = 1 mole of U =  $6.02 \times 10^{23}$  atoms of U.

Since equal number of gram atoms(moles) of different elements contain equal number of atoms.

Hence, 1 mole (23g ) of sodium and 1 mole (238)g of uranium contain equal number of atoms ,

i , e ,  $6.02 \times 10^{23}$  atoms.

# Short Questions Answers Of First Year Chemistry

**LM Reder**



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Teaching First Year Chemistry Joseph Otto Frank, 1927

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appropriate to exploring such transitions      The London Medical Record ,1885      The Journal of Education ,1892

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