SI Chemistry

Chemical Reactions Lab

<u>Purpose</u>: to become more familiar with different types of chemical reactions by conducting several reactions, noting the results, and discussing observations with colleagues.

<u>Procedure</u>: At each of the nine stations, there are directions for you to follow-<u>including</u> directions for waste disposal – FOLLOW ALL DIRECTIONS CAREFULLY!

You should visit only six of the nine stations. You MUST visit stations 6, 8 and 9. You will then choose one station from stations 1 and 7, one station from stations 2 and 5 and one station from stations 3 and 4. You do not need to move through the stations in any particular order. You MUST wear your safety goggles at ALL TIMES!

At each station, answer the following questions in your lab notebook or on your tablet:

- a. How did you know a chemical reaction took place?
- Formula of reactant(s)
- e. Formula of product(s)
- d. Type of chemical reaction
- e. Write the balanced equation for the reaction (including all physical states).

Station 1: Get a strip of magnesium ribbon (about 2.5 cm long). Holding it with forceps, ignite the ribbon in a Bunsen burner flame. Do not look directly at the burning magnesium!

Looking directly at the burning magnesium could damage your eyes. Hold the burning ribbon over a glass plate so that no burning magnesium lands on the lab bench.

Word equation: Magnesium reacts with oxygen to produce magnesium oxide.

Station 2: Place 10 drops of silver nitrate into a small test tube. Add 10 drops of sodium chloride to the same test tube.

Word equation: Aqueous silver nitrate reacts with aqueous sodium chloride to produce solid silver chloride and aqueous sodium nitrate.

Si Chemistry Chemical Reaction Lab Introduction

Daniel R. Palleros

Si Chemistry Chemical Reaction Lab Introduction:

Air Force Research Resumés , Electrochemical Sensors Giuseppe Maruccio, Jagriti Narang, 2022-02-01
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ERDA Energy Research Abstracts United States. Energy Research and Development Administration, 1977 Catalog of **Technical Reports** United States. Dept. of Commerce. Office of Technical Services, **ERDA Energy Research Abstracts** United States. Energy Research and Development Administration. Technical Information Center, 1977 Chemical Abstracts Experimental Organic Chemistry Daniel R. Palleros, 2000-02-04 This cutting edge lab manual takes a multiscale ,1909 approach presenting both micro semi micro and macroscale techniques. The manual is easy to navigate with all relevant techniques found as they are needed Cutting edge subjects such as HPLC bioorganic chemistry multistep synthesis and more are presented in a clear and engaging fashion Scientific and Technical Aerospace Reports ,1995-05 Energy Research Abstracts .1993 Technical Abstract Bulletin .1964 U.S. Government Research Reports ,1964 INIS Atomindex ,1986 **Nuclear Science Abstracts** ,1974 OTS Selective Bibliography United States. Dept. of Commerce. Office of Technical Services. Paperbacks in Print ,1974 Educational Media Index. [Complete Series] ,1964 Medical Books and Serials in Print, 1979 R. R. Bowker LLC, 1979-05 Japanese Technical Abstracts ,1987

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Table of Contents Si Chemistry Chemical Reaction Lab Introduction

- 1. Understanding the eBook Si Chemistry Chemical Reaction Lab Introduction
 - The Rise of Digital Reading Si Chemistry Chemical Reaction Lab Introduction
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Si Chemistry Chemical Reaction Lab Introduction
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Si Chemistry Chemical Reaction Lab Introduction
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Si Chemistry Chemical Reaction Lab Introduction
 - Personalized Recommendations
 - Si Chemistry Chemical Reaction Lab Introduction User Reviews and Ratings
 - Si Chemistry Chemical Reaction Lab Introduction and Bestseller Lists
- 5. Accessing Si Chemistry Chemical Reaction Lab Introduction Free and Paid eBooks

- Si Chemistry Chemical Reaction Lab Introduction Public Domain eBooks
- Si Chemistry Chemical Reaction Lab Introduction eBook Subscription Services
- Si Chemistry Chemical Reaction Lab Introduction Budget-Friendly Options
- 6. Navigating Si Chemistry Chemical Reaction Lab Introduction eBook Formats
 - o ePub, PDF, MOBI, and More
 - Si Chemistry Chemical Reaction Lab Introduction Compatibility with Devices
 - Si Chemistry Chemical Reaction Lab Introduction Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Si Chemistry Chemical Reaction Lab Introduction
 - Highlighting and Note-Taking Si Chemistry Chemical Reaction Lab Introduction
 - Interactive Elements Si Chemistry Chemical Reaction Lab Introduction
- 8. Staying Engaged with Si Chemistry Chemical Reaction Lab Introduction
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Si Chemistry Chemical Reaction Lab Introduction
- 9. Balancing eBooks and Physical Books Si Chemistry Chemical Reaction Lab Introduction
 - Benefits of a Digital Library
 - o Creating a Diverse Reading Collection Si Chemistry Chemical Reaction Lab Introduction
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Si Chemistry Chemical Reaction Lab Introduction
 - Setting Reading Goals Si Chemistry Chemical Reaction Lab Introduction
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Si Chemistry Chemical Reaction Lab Introduction
 - Fact-Checking eBook Content of Si Chemistry Chemical Reaction Lab Introduction
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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