

**Directions:** Using the Reference Tables for Chemistry, locate the following information.

1. Name  $C_5H_{12}$ . **Pentane** Table P + R
2. Write the equation for the decay of Kr-85.  $^{85}\text{Kr} \rightarrow \beta^- + ^{85}\text{Rb}$  Table N
3. Explain how you know that  $\text{NaPO}_4$  is soluble in water but  $\text{NiCrO}_4$  is not. **Na group 1/ all chromates insoluble** Table F
4. What is the definition of STP, and give the values? **Standard Temperature (0°C or 273K) and Pressure (1 atm or 101.3 kPa)** Table A
5. Name, and give the formulas of the strongest and weakest bases. **NaOH /  $\text{NH}_3$**  Table L
6. Name  $C_2H_3O_2^-$  or  $\text{CH}_3\text{COO}^-$  **acetate ion** Table E
7. What is the solubility of sulfur dioxide at 40 degrees Celsius, in grams? **~5g** Table G
8. What is the freezing point of fluorine? **53 K** Table S
9. What are the units for the heat of fusion, and what do they mean? **J/g heat lost per gram of substance** Table B
10. What is the symbol for the mole? **mol** Table D
11. What is the vapor pressure of water at 75°C? **38 kPa** Table H
12. How much heat does it take to convert 20g of water to steam at 100°C? **45200 J (45.2 kJ)** I
13. What is the prefix for 1/1000 of a meter? **Milli-** Table C
14. What is the molecular formula of ammonia?  **$\text{NH}_3$**  Table L
15. What is the formula for the permanganate ion?  **$\text{MnO}_4^-$**  Table E
16. Name  $\text{CH}_3\text{COOH}$  **ethanoic acid (acetic acid)** Table R
17. Write the symbol for a positron?  **$\beta^+$**  Table O
18. What is the half-life of Pu-239?  **$2.4 \times 10^4$  years** Table N
19. Is the formation of water, from its elements endothermic or exothermic? **exothermic** Table I
20. What is the atomic mass of silver? **108 g/mol** P-Table
21. How much heat is released when LiBr dissolves in water? **-48.83 kJ** Table I
22. Give the names and formulas of the strongest and weakest acids. **HCl /  $\text{CH}_3\text{COOH}$**  Table K
23. What is the general formula for alkynes? What does it mean?  **$C_nH_{2n-2}$**  Table Q

# Reference Table Scavenger Hunt Name C5h12 Answers

**DJ Losen**



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## **Table of Contents Reference Table Scavenger Hunt Name C5h12 Answers**

1. Understanding the eBook Reference Table Scavenger Hunt Name C5h12 Answers
  - The Rise of Digital Reading Reference Table Scavenger Hunt Name C5h12 Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Reference Table Scavenger Hunt Name C5h12 Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Reference Table Scavenger Hunt Name C5h12 Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Reference Table Scavenger Hunt Name C5h12 Answers
  - Personalized Recommendations
  - Reference Table Scavenger Hunt Name C5h12 Answers User Reviews and Ratings
  - Reference Table Scavenger Hunt Name C5h12 Answers and Bestseller Lists
5. Accessing Reference Table Scavenger Hunt Name C5h12 Answers Free and Paid eBooks
  - Reference Table Scavenger Hunt Name C5h12 Answers Public Domain eBooks
  - Reference Table Scavenger Hunt Name C5h12 Answers eBook Subscription Services
  - Reference Table Scavenger Hunt Name C5h12 Answers Budget-Friendly Options
6. Navigating Reference Table Scavenger Hunt Name C5h12 Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Reference Table Scavenger Hunt Name C5h12 Answers Compatibility with Devices
  - Reference Table Scavenger Hunt Name C5h12 Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Reference Table Scavenger Hunt Name C5h12 Answers
  - Highlighting and Note-Taking Reference Table Scavenger Hunt Name C5h12 Answers
  - Interactive Elements Reference Table Scavenger Hunt Name C5h12 Answers

8. Staying Engaged with Reference Table Scavenger Hunt Name C5h12 Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Reference Table Scavenger Hunt Name C5h12 Answers
9. Balancing eBooks and Physical Books Reference Table Scavenger Hunt Name C5h12 Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Reference Table Scavenger Hunt Name C5h12 Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Reference Table Scavenger Hunt Name C5h12 Answers
  - Setting Reading Goals Reference Table Scavenger Hunt Name C5h12 Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Reference Table Scavenger Hunt Name C5h12 Answers
  - Fact-Checking eBook Content of Reference Table Scavenger Hunt Name C5h12 Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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