#### Reduction-Oxidation Reactions Worksheet

# Part I: Identifying parts of Simple Red-ox reactions

For the equations below, all in aqueous solution:

- 1) assign oxidation numbers to each atom
- identify the reactant oxidized,
- 3) the reactant reduced,
- 4) the oxidizing agent,
- the reducing agent,
- write the balanced oxidation and reduction half reactions.

Example: oxidized reduced

Mg° + Br₂° → Mg\*² Br¹²

Reducing agent oxidizing agent

Oxidation half reaction: Mg° → Mg\*² + 2 e°

Reduction half reaction: 2 e° + Br₂° → 2Br°

Fe + Zn<sup>+2</sup> → Fe<sup>+2</sup> + Zn

Al + Fe<sup>+2</sup> → Al<sup>+3</sup> + Fe

3. Cu + AgNO<sub>3</sub>  $\rightarrow$  Cu(NO<sub>3</sub>)<sub>2</sub> + Ag

SiCl<sub>4</sub> (I) + Mg (s) → MgCl<sub>2</sub> (s) + Si (s)

# **Redox Reaction Exercise And Answers**

**Disha Experts** 

#### **Redox Reaction Exercise And Answers:**

Ionic Equilibria in Analytical Chemistry Jean-Louis Burgot, 2012-03-30 This book of general analytical chemistry as opposed to instrumental analysis or separation methods in aqueous solutions is focuses on fundamentals which is an area too often overlooked in the literature Explanations abound of the chemical and physical principles of different operations of chemical analysis in aqueous solutions Once these principle are firmly established numerous examples of applications are also given Basic Chemistry Concepts and Exercises John Kenkel, 2010-10-04 Chemistry can be a daunting subject for the uninitiated and all too often introductory textbooks do little to make students feel at ease with the complex subject matter Basic Chemistry Concepts and Exercises brings the wisdom of John Kenkel s more than 35 years of teaching experience to communicate the fundamentals of chemistry in a practical d A Guided Approach to Learning Chemistry Mailoo Selvaratnam, M. Selvaratnam, 1998 Stress is laid on the intellectual skills and strategies needed for learning and applying knowledge effectively in this foundation text Dr Selvaratnam sets out these strategies before focusing in on chemistry

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