

## REDOX REACTION

### Solved Examples

#### Example - 6 :



Base                  Acid

**Sol.** Valency factor of base = 1

Here, two molecule of NaOH replaced  $2\text{H}^+$  ion from the  $\text{H}_2\text{SO}_4$ . Therefore, each molecule of NaOH replaced only one  $\text{H}^+$  ion of acid, so v.f. = 1.

○ v. f. for acid is the number of  $\text{OH}^-$  replaced from the base by each molecule of acid

### Solved Examples

#### Example - 7 :



Base                  Acid

**Sol.** Valency factor of acid = 1

Here, one of molecule of  $\text{H}_2\text{SO}_4$  replaced one  $\text{OH}^-$  from NaOH. Therefore, valency factor for  $\text{H}_2\text{SO}_4$  is one

$$\therefore \text{Eq. wt. of } \text{H}_2\text{SO}_4 = \frac{\text{Mol. wt.}}{1}$$

○ Salts :

(a) In non-reacting condition

○ Valency factor = Total number of positive charge or negative charge present in the compound.

### Solved Examples

#### Example - 8 :

		$\text{Na}_2\text{CO}_3$	$\text{Fe}_2(\text{SO}_4)_3$	$\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$
<b>Sol.</b>	v.f.	2	$2 \times 3 = 6$	2
	Eq. wt.	$\frac{M}{2}$	$\frac{M}{6}$	$\frac{M}{2}$

(b) In reacting condition

### Solved Examples

#### Example - 9 :



Base                  Acid

**Sol.** It is an acid base reaction, therefore valency factor for  $\text{Na}_2\text{CO}_3$  is one while in non-reacting condition, it will be two.

(c) Equivalent weight of oxidising / reducing agents in a redox reaction

In case of redox change, v.f. = Total change in oxidation number per molecule.

# Redox Reactions Chapter Assessment Answers

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## **Redox Reactions Chapter Assessment Answers:**

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the very beginning through reflective teaching The volume includes detailed descriptions of class room experiments and structural models to cure and to prevent these misconceptions      **Chemistry** John Olmsted,Greg Williams,Robert C. Burk,2020 Chemistry 4th Edition is an introductory general chemistry text designed specifically with Canadian professors and students in mind A reorganized Table of Contents and inclusion of SI units IUPAC standards and Canadian content designed to engage and motivate readers and distinguish this text from other offerings It more accurately reflects the curriculum of most Canadian institutions Chemistry is sufficiently rigorous while engaging and retaining student interest through its accessible language and clear problem solving program without an excess of material and redundancy      **Sif: Chemistry 5na Tb** J. G. R. Briggs,2009      *Cracking the SAT Subject Test in Chemistry, 16th Edition* The Princeton Review,2018-02-13 EVERYTHING YOU NEED TO HELP SCORE A PERFECT 800 Equip yourself to ace the SAT Subject Test in Chemistry with The Princeton Review s comprehensive study guide including 3 full length practice tests thorough reviews of key chemistry topics and targeted strategies for every question type We don t have to tell you how tough SAT Chemistry is or how helpful a stellar exam score can be for your chances of getting into your top choice college Written by the experts at The Princeton Review Cracking the SAT Subject Test in Chemistry arms you to take on the test and achieve your highest score Techniques That Actually Work Tried and true strategies to help you avoid traps and beat the test Tips for pacing yourself and guessing logically Essential tactics to help you work smarter not harder Everything You Need to Know for a High Score Expert subject reviews for every test topic Up to date information on the SAT Subject Test in Chemistry Score conversion tables for accurate self assessment Practice Your Way to Perfection 3 full length practice tests with detailed answer explanations Hands on experience with all three question types in each content chapter Complete study sheet of core formulas and terms This eBook edition has been optimized for on screen learning with cross linked questions answers and explanations      **Neet Chapter-Wise & Topic-Wise Solved Papers: Chemistry (2005-2022) With 5 Mock Test** Dr. K. G. Ojha,2022-06-02 The knowledge of Chemistry helps you to understand the world around you From food to pharmaceuticals Chemistry plays a huge role in making informed decisions Therefore to brush up your intellect we present the NEET Chapterwise and Topicwise Chemistry Solved Papers 2005 2021 which is designed to provide a simplified yet systematic understanding to ace the examination The Study Material is strictly based on NCERT Latest Exam Solved Paper is included The Concepts are explained in depth Chapters are compiled with Previous Years Questions Answers to Questions included with Explanations Presence of accurate Figures throughout Five sets of Mock Tests are also included at the end This title focuses on an all inclusive preparations providing the aspirants to learn revise test and gauge their progress against the examination level The Book contains the following units Unit I Physical Chemistry I Unit II Physical Chemistry II Unit III Organic Chemistry I Unit IV Organic Chemistry II Unit V Inorganic Chemistry I Unit VI Inorganic Chemistry II      A text book of Pharmaceutical inorganic chemistry for 1st year Pharm.D (as per PCI,New Delhi Norms) Dr. M.P. Bhagat,2022-12-28

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