

Regents Chemistry
Unit 8 Thermochemistry Practice Test

- At which temperature do both the solid and the liquid phase of water exist?
(1) 273 °C (3) 0 °C
(2) 100 °C (4) 373 °C
- What is the freezing point of sodium?
(1) 371 °C (3) -98 °C
(2) 644 °C (4) 98 °C
- A 15.75-g piece of iron absorbs 1086.75 joules of heat energy, and its temperature changes from 25 °C to 175 °C. What is the specific heat capacity of iron?
(1) 1.0 J/g °C (3) 0.46 J/g °C
(2) 4.18 J/g °C (4) 2.76 J/g °C
- Which statement concerning propanone and water at 50 °C is true?
(1) Propanone has a lower vapor pressure and weaker intermolecular forces than water.
(2) Propanone has a lower vapor pressure and stronger intermolecular forces than water.
(3) Propanone has a higher vapor pressure and weaker intermolecular forces than water.
(4) Propanone has a higher vapor pressure and stronger intermolecular forces than water.
- Which phase change is an exothermic process?
(1) $\text{CO}_2 (\text{s}) \rightarrow \text{CO}_2 (\text{g})$ (2) $\text{Cu} (\text{s}) \rightarrow \text{Cu} (\text{l})$
(3) $\text{NH}_3 (\text{l}) \rightarrow \text{NH}_3 (\text{g})$ (4) $\text{Hg} (\text{l}) \rightarrow \text{Hg} (\text{s})$
- What is the vapor pressure of water at 105 °C?
(1) 100 kPa (3) 101.3 kPa
(2) 110 kPa (4) 120 kPa
- As a substance changes from a liquid to a gas, the average distance between molecules
(1) decreases
(2) increases
(3) remains the same

- The graph below represents the uniform heating of a substance, starting with the substance as a solid below its melting point.



Which line segment represents an increase in average kinetic energy and *no* change in potential energy?

- (1) BC (3) ED
(2) AB (4) DE
- How many joules are required to melt 100. grams of water?
(1) 226,000 kJ (2) 22,600 J
(3) 33,400 kJ (4) 33,400 J
- What occurs when a 35-gram aluminum cube at 100. °C is placed in 90. grams of water at 25 °C in an insulated cup?
(1) Heat is transferred from the aluminum to the water, and the temperature of the water decreases.
(2) Heat is transferred from the aluminum to the water, and the temperature of the water increases.
(3) Heat is transferred from the water to the aluminum, and the temperature of the water decreases.
(4) Heat is transferred from the water to the aluminum, and the temperature of the water increases.
- What is the boiling point of propanone if the pressure on its surface is 48 kilopascals?
(1) 25 °C (3) 35 °C
(2) 30 °C (4) 40 °C

Regents Questions On Thermochemistry

Richard Bailey



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Table of Contents Regents Questions On Thermochemistry

1. Understanding the eBook Regents Questions On Thermochemistry
 - The Rise of Digital Reading Regents Questions On Thermochemistry
 - Advantages of eBooks Over Traditional Books
2. Identifying Regents Questions On Thermochemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Regents Questions On Thermochemistry
 - User-Friendly Interface
4. Exploring eBook Recommendations from Regents Questions On Thermochemistry
 - Personalized Recommendations
 - Regents Questions On Thermochemistry User Reviews and Ratings
 - Regents Questions On Thermochemistry and Bestseller Lists
5. Accessing Regents Questions On Thermochemistry Free and Paid eBooks
 - Regents Questions On Thermochemistry Public Domain eBooks
 - Regents Questions On Thermochemistry eBook Subscription Services
 - Regents Questions On Thermochemistry Budget-Friendly Options

6. Navigating Regents Questions On Thermochemistry eBook Formats
 - ePub, PDF, MOBI, and More
 - Regents Questions On Thermochemistry Compatibility with Devices
 - Regents Questions On Thermochemistry Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Regents Questions On Thermochemistry
 - Highlighting and Note-Taking Regents Questions On Thermochemistry
 - Interactive Elements Regents Questions On Thermochemistry
8. Staying Engaged with Regents Questions On Thermochemistry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Regents Questions On Thermochemistry
9. Balancing eBooks and Physical Books Regents Questions On Thermochemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Regents Questions On Thermochemistry
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Regents Questions On Thermochemistry
 - Setting Reading Goals Regents Questions On Thermochemistry
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Regents Questions On Thermochemistry
 - Fact-Checking eBook Content of Regents Questions On Thermochemistry
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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