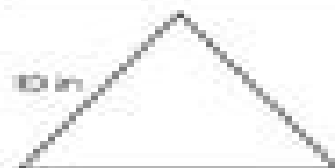


Area of Regular Figuresnotes continued

When you do not have the apothem or a side length:

- Create a right triangle by drawing the apothem and radius.
- Find the angles of the right triangle.
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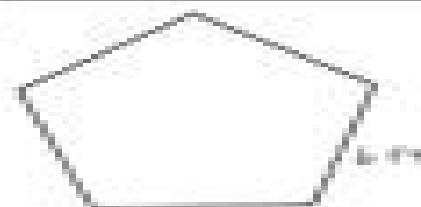
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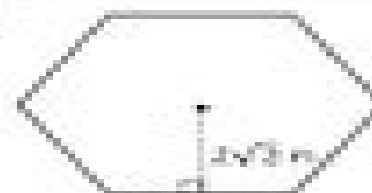
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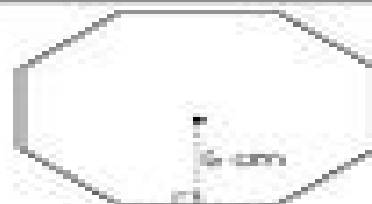
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9.



10.



11.



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R Barnett

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buffers typically consist of an acid base pair with the acid and base differing by the presence or absence of a proton a conjugate acid base pair for instance one of the buffers that maintain the ph of human blood involves carbonic acid H_2CO_3 and its conjugate base the bicarbonate ion HCO_3^-

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acid base titration difference between acid and base acids and bases definition an acid is any hydrogen containing substance that is capable of donating a proton hydrogen ion to another substance a base is a molecule or ion able to accept a hydrogen ion from an acid acidic substances are usually identified by their sour taste

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arrhenius's definition of acids and bases the earliest definition of acids and bases is arrhenius's definition which states that an acid is a substance that forms hydrogen ions H^+ when dissolved in water and a base is a substance that forms hydroxide ions OH^- when dissolved in water

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the reaction of an acid and a base is called a neutralization reaction although acids and bases have their own unique chemistries the acid and base cancel each other's chemistry to produce a rather innocuous substance water in fact the general reaction between an acid and a base is

$$\text{acid} + \text{base} \rightarrow \text{water} + \text{salt}$$

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