

Answer Key

Guided Reading 6.1

Observing Chemical Change Guided Reading and Study Use Target Reading Skills Sample questions and answers: Question What are physical properties of matter? Answer Physical properties are characteristics that can be observed without changing one substance into another. Question What is the evidence for chemical reactions? Answer The evidence for chemical reactions is the formation of new substances and changes in energy.

1. Matter is anything that has mass and takes up space.
2. chemistry
3. a. Physical b. Only by changing one substance into another
4. False
5. a, b, d
6. chemical reaction, or chemical change
7. They break and form new bonds, producing new substances.
8. Formation of new substances and changes in energy
9. The change in color is an indication of a change in properties, and a change in properties is a sign that a chemical reaction occurred.
10. precipitate
11. It might indicate a chemical reaction, because the bubbles are gas, and the presence of a gas indicates a new substance may have been formed.
12. True
13. A change in temperature reflects a change in energy.
14. False
15. a. Energy is absorbed b. Exothermic

6.2

Describing Chemical Reactions Guided Reading and Study Use Target Reading Skills a. A chemical equation is a short, easy way to show a chemical reaction. b. A substance you have at the beginning of a reaction is a reactant. c. A new substance produced in a reaction is a product. d. Conservation of mass means that during a chemical reaction, matter is not created or destroyed. e. In an open system, matter can enter from or escape to the surroundings. f. In a closed system, matter is not allowed to enter or leave. g. A coefficient is a number in a chemical equation telling you how many atoms or molecules of a

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Academy of Engineering, National
Science Resources Center of the
National Academy of
Sciences, Institute of Medicine**

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and Multidisciplinary and Applied Science They are also grouped by type core materials supplementary units and science activity books Each annotation of curriculum material includes a recommended grade level a description of the activities involved and of what students can be expected to learn a list of accompanying materials a reading level and ordering information The curriculum materials included in this book were selected by panels of teachers and scientists using evaluation criteria developed for the guide The criteria reflect and incorporate goals and principles of the National Science Education Standards The annotations designate the specific content standards on which these curriculum pieces focus In addition to the curriculum chapters the guide contains six chapters of diverse resources that are directly relevant to middle school science Among these is a chapter on educational software and multimedia programs chapters on books about science and teaching directories and guides to science trade books and periodicals for teachers and students Another section features institutional resources One chapter lists about 600 science centers museums and zoos where teachers can take middle school students for interactive science experiences Another chapter describes nearly 140 professional associations and U S government agencies that offer resources and assistance Authoritative extensive and thoroughly indexed and the only guide of its kind Resources for Teaching Middle School Science will be the most used book on the shelf for science teachers school administrators teacher trainers science curriculum specialists advocates of hands on science teaching and concerned parents Chemical Interactions ,2005 **Modules** McDougal Littell Incorporated,2005 **Investigations in**

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