

Polyatomic Ions WS

Ammonium (NH_4)¹⁺
 Carbonate (CO_3)²⁻
 Chromate (CrO_4)²⁻

Dichromate (Cr_2O_7)²⁻
 Hydroxide (OH)¹⁻
 Nitrate (NO_3)¹⁻

Nitrite (NO_2)¹⁻
 Sulfate (SO_4)²⁻
 Sulfite (SO_3)²⁻

Phosphate (PO_4)³⁻

Write the NAME of each of the following compounds.

1. $(\text{NH}_4)\text{Cl}$ Ammonium Chloride
2. $\text{Be}(\text{SO}_4)$ Beryllium Sulfate
3. $(\text{NH}_4)_3\text{N}$ Ammonium Nitride
4. MgCl_2 Magnesium Chloride
5. $\text{NH}_4(\text{NO}_3)$ Ammonium Nitrate
6. $\text{Sr}_3(\text{PO}_4)_2$ Strontium Phosphate
7. $\text{Zn}(\text{CrO}_4)_2$ Zinc Dichromate
8. $\text{K}_2(\text{Cr}_2\text{O}_7)$ Potassium Dichromate

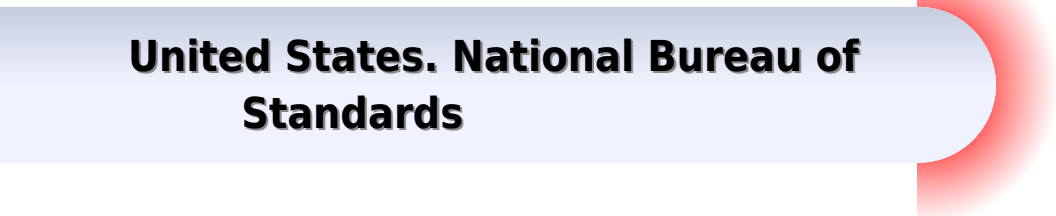
9. $\text{Ga}(\text{ClO}_3)_3$ Gallium Chlorate
10. $\text{Cu}(\text{OH})$ Copper I Hydroxide
11. $(\text{NH}_4)_3(\text{PO}_4)$ Ammonium Phosphate
12. $\text{Fe}(\text{SO}_4)$ Iron II Sulfate
13. $\text{Mg}(\text{NO}_2)_2$ Magnesium Nitrite
14. $(\text{NH}_4)\text{NO}_2$ Ammonium Nitrite
15. $\text{Na}_2(\text{Cr}_2\text{O}_7)$ Sodium Dichromate
16. $\text{Na}(\text{OH})$ Sodium Hydroxide

Write the CHEMICAL FORMULA for each of the given NAMES.....cross charges

- | | | | | |
|-------------------------|-------------------------------------|---|-------------------------------|-------------------|
| 17. calcium carbonate | $\text{Ca}^{2+} (\text{CO}_3)^{2-}$ | = | $\text{Ca}_2(\text{CO}_3)_2$ | = CaCO_3 |
| 18. barium nitrate | $\text{Ba}^{+2} \text{NO}_3^{-1}$ | = | $\text{Ba}(\text{NO}_3)$ | |
| 19. ammonium sulfate | $\text{NH}_4^{+1} \text{SO}_4^{-2}$ | = | $(\text{NH}_4)_2 \text{SO}_4$ | |
| 20. aluminum hydroxide | $\text{Al}^{+3} \text{OH}^{-1}$ | = | $\text{Al}(\text{OH})_3$ | |
| 21. calcium phosphate | $\text{Ca}^{+2} \text{PO}_4^{-3}$ | = | $\text{Ca}_3(\text{PO}_4)_2$ | |
| 22. cesium nitrate | $\text{Cs} \text{NO}_2^{-1}$ | = | CsNO_3 | |
| 23. sodium nitrite | $\text{Na}^{+1} \text{NO}_3^{-1}$ | = | NaNO_3 | |
| 24. calcium sulfate | $\text{Ca}^{+2} \text{SO}_4^{-2}$ | = | CaSO_4 | |
| 25. beryllium sulfate | $\text{Be} \text{SO}_4^{-2}$ | = | BeSO_4 | |
| 26. sodium carbonate | $\text{Na}^{+1} \text{CO}_3^{-2}$ | = | Na_2CO_3 | |
| 27. magnesium phosphate | $\text{Mg}^{+2} \text{PO}_4^{-3}$ | = | $\text{Mg}_3(\text{PO}_4)_2$ | |
| 28. calcium phosphate | $\text{Ca}^{+2} \text{PO}_4^{-3}$ | = | $\text{Ca}_3(\text{PO}_4)_2$ | |

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Polymer Surface Characterization Luigia Sabbatini, Elvira De Giglio, 2022-01-19 This fully updated edition provides a broad approach to the surface analysis of polymers being of high technological interest Modern analytical techniques potential applications and recent advances in instrumental apparatus are discussed The self consistent chapters are devoted to spectroscopic and microscopic techniques which represent powerful tools for the characterization of morphology and chemical physical mechanical properties of polymer surfaces interfaces and thin films Selection of techniques which can properly address very shallow depth of surfaces spanning from few angstroms to tens of nanometers Interaction of polymer surfaces with their surroundings is pointed out as a critical issue for specific applications Ion/Molecule Attachment Reactions: Mass Spectrometry Toshihiro Fujii, 2015-08-03 This book explores the mechanism of alkali metal ion molecule association reaction surveys the instrumental basis to study its kinetic and describes the instrumentation to the measurement of alkali metal ion affinities The applications of the ion complexation mechanism in the condensed phase in reaction to direct analysis MS are also covered Other topics include mechanism and reaction rate experimental and theoretical ion affinities applications of ion attachment reactions IAR to mass spectrometry such as alkali ion CIMS ion attachment MS and cationization mass spectrometry of ESI FAB FD LD MALDI and SIMS and topics of IAR based direct analysis mass spectrometry **Research in Progress**, 1983 Recent Studies in Atomic and Molecular Processes Arthur E. Kingston, 2012-12-06 Sir David Bates celebrated his seventieth birthday on the 18th November 1986 To mark this event a conference was held in the David Bates building at The Queen's University of Belfast on the 17th and 18th November 1986 At this conference ex students and colleagues of Sir David who are acknowledged world experts in their field of research gave in depth reviews of a particular area of atomic and molecular physics This book on the conference presents a unique account of recent studies in atomic and molecular processes in a wide range of research fields This volume is dedicated to Sir David by his friends as a token of their affection and respect It is hoped that it will provide a useful summary of current research in atomic and molecular physics and that it will also show the great contribution which Sir David made to atomic and molecular physics This conference was supported by the USAF European Office of Aerospace Research and Development who we wish to thank for their generous support Arthur E Kingston v CONTENTS Interstellar Cloud Chemistry Revisited 1 D R Bates Photo ionisation of Atomic Oxygen 29 M J Seaton The Formation of Complex Interstellar Molecules **Research in Progress** United States. Army Research Office, 1983 Vols for 1977 consist of two parts Chemistry biological sciences engineering sciences metallurgy and materials science issued in the spring and Physics electronics mathematics geosciences issued in the fall Journal of Physical Chemistry American Chemical Society, 1983 Mass Spectrometry R. A. W. Johnstone, 1975-12-31 Specialist Periodical Reports provide systematic and detailed review coverage of progress in the major areas of chemical research Written by experts in their specialist fields the series creates a unique service for the active

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Molecular Ions Joseph Berkowitz, Karl-Ontjes E. Groeneveld, 2012-12-06 The present volume contains contributions presented at the NATO Advanced Study Institute on Molecular Ions held on the island of Kos Greece from September 30 to October 10 1980 The meeting was attended by some 60 participants from 15 different countries It was the first meeting devoted exclusively to the topic of molecular ions Its vitality derived from bringing together experts and students from a wide variety of disciplines whose studies bear upon the structure of molecular ions The aim of the meeting was to assemble these scientists representing many countries in Europe and North America to discuss the advances and

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