

OCR ANSWER AND SOLUTIONS

1 How many electrons are in a $^{24}_{12}\text{Mg}^{2+}$ ion?

- A 10
- B 12
- C 14
- D 22

Your answer - ANSWER A

2 What is the formula of chromium(III) sulfate?

- A Cr_3SO_4
- B $\text{Cr}(\text{SO}_4)_3$
- C $\text{Cr}_2(\text{SO}_4)_3$
- D Cr_3SO_3

Your answer - ANSWER C

3 Which molecule is non-polar?

- A SF_6
- B H_2S
- C PF_3
- D NH_3

Your answer - ANSWER A

4 Which row is correct?

Highest pH when added to water Most reactive halogen

- A MgO F_2
- B MgO I_2
- C BaO F_2
- D BaO I_2 - ANSWER C

5 Which equation represents a redox reaction?

- A $\text{Mg} + 2\text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2$
- B $\text{MgO} + 2\text{HCl} \rightarrow \text{H}_2\text{O} + \text{MgCl}_2$
- C $\text{MgCO}_3 + 2\text{HCl} \rightarrow \text{CO}_2 + \text{H}_2\text{O} + \text{MgCl}_2$
- D $\text{Mg}(\text{OH})_2 + 2\text{HCl} \rightarrow \text{MgCl}_2 + 2\text{H}_2\text{O}$ - ANSWER A

6 This question is about trends in the periodic table.

Which trend is correct?

- A melting point decreases from lithium to carbon
- B boiling point decreases from fluorine to iodine
- C first ionisation energy decreases from lithium to caesium
- D first ionisation energy increases from nitrogen to oxygen - ANSWER C

7 A sample of a compound M contains 1.46 g of carbon, 0.482 g of hydrogen and 1.69 g of nitrogen.

What is the empirical formula of M?

- A CH_2N
- B C_4HN_4
- C CH_4N

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KJ Lindholm-Leary



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