

TYBSC SEM VI PHYSICAL CHEMISTRY (MODEL MCQ's)

1	Which of the following is a uni-univalent electrolyte? a) KCl b) $\text{CuCl}_2$ c) $\text{AlCl}_3$ d) $\text{MgCl}_2$
2	The ionic strength of a solution of molality $m_i$ and valency of the ion $Z_i$ is ____ a) $\mu = \frac{1}{2} m_i Z_i^2$ b) $\mu = \sum m_i Z_i^2$ c) $\mu = \frac{1}{2} \sum Z_i^2$ d) $\mu = \frac{1}{2} \sum m_i Z_i^2$
3	Which of the following is an example of electrolyte concentration cell with transference?  a) $\text{Cu}   \text{CuSO}_4    \text{AgNO}_3   \text{Ag}$ b) $\text{Pt}, \text{H}_{2(g)}   \text{HCl}_{(aq)}   \text{AgCl}_{(s)}   \text{Ag}$ c) $\text{Pt}, \text{H}_{2(g)}   \text{HCl}_{(aq)}(a_1)   \text{AgNO}_3(a_2)   \text{Ag}$ d) $\text{Ag} - \text{AgCl}   \text{NaCl}(a_1)   \text{NaCl}(a_2)   \text{AgCl} + \text{Ag}$
4	A concentration cell which has salt bridge in its representation is ____ a) electrolyte concentration cell with transference b) Electrode concentration cell c) Electrolyte concentration cell without transference d) Gas concentration cell
5	The total emf of the electrolyte concentration cell with transference containing the liquid junction potential is given by..... a) $E_j = E_{\text{oxidation}} + E_{\text{reduction}}$ b) $E_{\text{cell}} = E_{\text{oxidation}} + E_{\text{reduction}} + E_j$ c) $E_{\text{cell}} = E_{\text{oxidation}} - E_{\text{reduction}} - E_j$ d) $E_j = E_{\text{oxidation}} - E_{\text{reduction}}$
6	The activity 'a' is related to the activity coefficient ' $\gamma$ ' by the equation____ a) $a = \frac{m}{\gamma}$ b) $\gamma = a - m$ c) $a = m + \gamma$ d) $a = m\gamma$

# Physical Chemistry Multiple Choice Questions

**Princeton Review (Firm)**



## Physical Chemistry Multiple Choice Questions:

**Multiple Choice Questions in Physical Chemistry for Undergraduate Students** Dr Kusum Baser, 2022-08-25

Objective questions are very much useful for practice of particular topic Multiple choice questions are important to practice any topic to solve objective questions very deep study and knowledge of specific topic is needed This book helps the students to do practice in such a way so they can do preparation for competitive exams This book covers important topics of physical chemistry for undergraduate students **Physical Chemistry: A Level Multiple Choice Questions** Margaret Marion

Langley Oblitas, 1970-01-01 Physical Chemistry M.M.L. Oblitas, 1977 *Physical Chemistry* M.M.L. Oblitas, 1971 **The Pearson Guide to Physical Chemistry for the IIT JEE** Singhal, **Pharmaceutical Physical Chemistry: Theory and Practices** S. K. Bhasin, Physical chemistry is a compulsory paper offered to all the students of pharmacy There is a dearth of good books that exclusively cover the syllabi of physical chemistry offered to pharmacy courses Pharmaceutical Physical Chemistry Theory and Practices has been designed considering their requirements laid down by AICTE and other premier institutes universities Apart from the theory 20 most common laboratory experiments have been included to make this book a unique offering to the students of pharmacy **A Textbook of Physical Chemistry, 6th Edition** Sharma K.K. & Sharma L.K., 2016 A Textbook of Physical Chemistry *Atkins' Physical Chemistry* Peter William Atkins, Julio De Paula, James Keeler, 2023 Revised edition of Atkins Physical chemistry Peter Atkins Julio de Paula James Keeler Eleventh edition 2018

*Elements of Physical Chemistry* Peter Atkins, Julio de Paula, 2013 Elements of Physical Chemistry has been carefully crafted to help students increase their confidence when using physics and mathematics to answer fundamental questions about the structure of molecules how chemical reactions take place and why materials behave the way they do Working Method Approach for Introductory Physical Chemistry Calculations Brian J Hathaway, Clair Murphy, Brian Murphy, 2019-05-16 A Working Method Approach for Introductory Physical Chemistry Calculations is a concise inexpensive introduction to first year chemistry that is aimed at students who are weak in chemistry or have no chemistry on entry to university Such students usually find physical chemistry the most difficult part of the chemistry course and within this section numerical problem solving is an additional difficulty The text should also be invaluable to first year intending chemists This text provides an introduction to physical chemistry and the gas laws followed by chapters on thermodynamics chemical equilibrium electrochemistry and chemical kinetics Each section involves a brief introduction followed by a representative examination question which is broken down into a proposed working method Both short multiple choice questions and related full examination type questions are included This book will prove invaluable to students who need encouragement in a logical approach to problem solving in physical chemistry teaching them to think for themselves when faced with a problem

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*Physical chemistry 2* Alyn G. McFarland, Nora Henry, 2016-04-04 Exam Board AQA Level A level Subject Chemistry First Teaching September 2016 First Exam June 2017 Written by experienced examiners Alyn McFarland and Nora Henry this Student Guide for Chemistry Helps students identify what they need to know with a concise summary of the topics examined in the AS and A level specifications Consolidates understanding with tips and knowledge check questions Provides opportunities to improve exam technique with sample answers to exam style questions Develops independent learning and research skills Provides the content for generating individual revision notes      AQA AS/A Level Year 1 Chemistry Student Guide: Physical chemistry 1 Alyn G. McFarland, Nora Henry, 2015-10-09 Exam Board AQA Level AS A level Subject Chemistry First Teaching September 2015 First Exam June 2016 Written by experienced examiners Alyn McFarland and Nora Henry this Student Guide for Chemistry Helps you identify what you need to know with a concise summary of the topics examined in the AS and A level specifications Consolidates understanding with tips and knowledge check questions Provides opportunities to improve exam technique with sample answers to exam style questions Develops independent learning and research skills Provides the content for generating individual revision notes      **Essentials of Physical Chemistry 28th Edition** Bahl Arun/ Bahl B.S. & Tuli G.D., 2022-03 Essentials of Physical Chemistry is a classic textbook on the subject explaining fundamentals concepts with discussions illustrations and exercises With clear explanation systematic presentation and scientific accuracy the book not only helps the students clear misconceptions about the basic concepts but also enhances students ability to analyse and systematically solve problems This bestseller is primarily designed for B Sc students and would equally be useful for the aspirants of medical and engineering entrance examinations      Elements of Physical Chemistry Peter William Atkins, Julio De Paula, 2017 This revision of the introductory textbook of physical chemistry has been designed to broaden its appeal particularly to students with an interest in biological applications      Multiple Choice Questions in A-level Chemistry B. P. Bourne, 1969      **Physical Chemistry of Polymers** Sebastian Seiffert, 2023-01-30 This book introduces the concepts of physical chemistry of polymers in a format targeted for a blended learning approach It provides a basis to bridge polymer chemistry which targets microscopic chain structures and polymer engineering which targets macroscopic material properties and functions Topics covered are single chain statistics multi chain interactions and chain dynamics both from a viewpoint of structure properties mostly mechanical ones and their interrelation In all that the author encourages the reader to think conceptually Explains complex facts through simplifying models diagrams and illustrations Accessible to chemists chemical engineers materials scientists and physicists Tailored content for an interactive blended learning format      *Cracking the GRE Chemistry Test*, 2002 The Princeton Review realizes that acing the GRE Chemistry Test is very different from getting straight A s in school We don t try to teach you everything there is to know about chemistry only the techniques you ll need to score higher on the test There s a big difference In *Cracking the GRE Chemistry Test* we ll teach you how to think like the test writers and Eliminate answer choices that look right but are planted

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