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EDUCATION
REPUBLIC OF SOUTH AFRICA

**GAUTENG DEPARTMENT OF EDUCATION
PROVINCIAL EXAMINATION
JUNE 2017
GRADE 11**

**PHYSICAL SCIENCES
PAPER 2**

CHEMISTRY

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web bismuth has a molar mass of one mole of the compound would result in two moles of bismuth a total of 417 96g tellurium

has a molar mass of one mole of the compound would result in three moles of tellurium a total of 382 8g add the mass of

bismuth and the mass of tellurium per mole to find the total molar mass divide the mass of bismuth

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web in molecular compounds the atom binds each other through covalent bonds in salts it is held together with ionic bonds

these are the two types of bonds out of which every compound is made of 2 example of compounds example of compounds

includes water h₂o hydrogen peroxide h₂o₂ etc

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web the substances mentioned above exemplify the two basic types of chemical compounds molecular and ionic methane and

water are composed of molecules that is they are molecular compounds sodium chloride on the other hand contains ions it is

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web molecular compounds many compounds do not contain ions but instead consist solely of discrete neutral molecules these

molecular compounds covalent compounds result when atoms share rather than transfer gain or lose electrons covalent

bonding is an important and extensive concept in chemistry and it will be treated in considerable

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it is ionic write the symbols for the ions involved a nf 3 b ba o c nh 4 2 co 3 d sr h 2 po 4 2 e ibr f na 2 o answer a answer b

answer c answer d answer e answer f problem 4 3 1 4 4 3 1 4

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web jul 25 2022 write the name for each molecular compound co h 2s sf 6 n 2o 2 strategy first you have to know how to name a molecular compound to name a compound its the prefix name of first element prefix and name of the second element with the suffix ide the prefixes are as follows mono 1 di 2 tri 3 tetra 4

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web molecular compounds are inorganic compounds that take the form of discrete molecules examples include such familiar substances as water h₂o h 2 o and carbon dioxide co₂ co 2 these compounds are very different from ionic compounds like sodium chloride nacl nacl

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