

# Periodic Table Basics

**Step 1:** Complete the square for each element by filling in the atomic number, name, & atomic mass.

**Step 2:** Determine the number of protons, neutrons, and electrons in an atom of each element.

**Step 3:** Identify if the element is a solid, liquid, or gas at room temperature.

**Step 4:** Give the melting (M.P.) and boiling points (B.P.) in degrees Celsius.

**Step 5:** List at least three physical or chemical properties for each element.

**Step 6:** List at least three uses for each element.

**Step 7:** Draw a Bohr diagram and Lewis Structure to show the arrangement of electrons and the number of valence electrons.

**Step 8:** Use the following colors to shade in the square for each element. You should **ONLY** color the small square in the upper left-hand corner and not the entire card.

Green = Li & Na  
Orange = B & Al


Pink = O & S  
Red = C & Si

Blue = Be & Mg  
Tan = N & P

Purple = F & Cl  
Yellow = He, Ne, & Ar

**Step 9:** Cut the cards apart and arrange according to atomic number in the pattern shown. Once you have the cards arranged in the correct order, glue them to a large sheet of construction paper.

1							2
3	4	5	6	7	8	9	10
11	12	13	14	15	16	17	18

① <div style="text-align: center; font-size: 2em;">B</div> <hr/> <hr/>	P= ② N= _____ E= _____	③ O S O L O G
	M.P. = _____ B.P. = ④ _____	
Properties ⑤  Uses ⑥  Bohr Diagram      Lewis Structure <div style="display: flex; align-items: center; justify-content: space-around;">  <div>⑦</div> <div style="font-size: 1.5em;">B</div> </div>		

Done? Answer the questions on the worksheet using the information on your Periodic Table!

# Physical Science If8767 Page 64 Answers

**David Frank, Michael Wyssession, Sophia  
Yancopoulos**



## **Physical Science If8767 Page 64 Answers:**

Physical Science Arthur Godman,1976      Introduction Physical Science Puri,1999-05-01      Physical Science Robert H. Marshall,Donald H. Jacobs,1992      **Macmillan Physical Science** Denise Eby,Robert B. Horton,1988      **Physical Science** National Learning Corporation,2018 The DSST Physical Science Passbook R prepares candidates for the DSST exam which enables schools to award credit for knowledge acquired outside the normal classroom environment It provides a series of informational texts as well as hundreds of questions and answers in the areas that will likely be covered on your upcoming exam including but not limited to physics electricity and magnetism Glossyr chemical reactions atomic structure and more

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Holt Physical Science William L. Ramsey,1978      **Physical Science** National Geographic Society (U.S.),Glencoe/McGraw-Hill,2002\*      **Oklahoma Physical Science** ,2016      **An Introduction to Physical Science** Jerry Wilson,Charles Higgins,James Shipman,2013      Study Guide to Physical Science Leon Horn,Benjamin du Preez Brink,Ronald Colin Jones,1989      **Introduction to Physical Science** Shipman,2015

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