



NATIONAL
SENIOR CERTIFICATE

GRADE 11

NOVEMBER 2020

PHYSICAL SCIENCES P2
(CHEMISTRY)
EXEMPLAR

MARKS: 100
TIME: 1 hour



The question paper consists of 10 pages, including 4 extra sheets.
Use permanent marker for writing and drawing.

QUESTION 3

PHYSICAL SCIENCES PAPER 2

INTERMOLECULAR FORCES

GRADE 11

NOVEMBER 2020

Examiner's use: _____ Physical Sciences P2

QUESTION 3 starts on a NEW page.

3.1 The boiling points in the table below were obtained during an investigation carried out to measure the boiling points of pure 1-hexanol. The table also lists the ΔH_{vap} and ΔH_{fusion} for 1-hexanol in kJ mol⁻¹ and 1 mol of 1-hexanol.

Compound	Molecular mass (M _r)	Boiling point (°C)
A	102	16,2
B	102	21

- 3.1.1 Define the term boiling point. (2)
- 3.1.2 Is pure 1-hexanol only composed of A and B or of A, B and C or of others as well? (2)
- 3.1.3 Name the type of intermolecular force that exists between molecules of both compounds A and B and is the prime reason of their volatility. (2)
- 3.1.4 What (100% of the) compound A or B is less (1-hexanol) volatile than (propyl) 1-hexanol? (2)
- 3.1.5 Which compound (A or B) will have the higher vapour pressure? (2)
- 3.1.6 Give a reason for your answer. (2)
- 3.1.7 Which compound (A or B) will have the higher surface tension? (2)
- 3.1.8 Give a reason for the answer to 3.1.7 relating to the space the table. (2)

Compound C (M_r 102) will show in the table, has a smaller molecular mass than both compounds A and B but has a relatively higher boiling point of 16,2 °C.

- 3.1.8 Explain why the boiling point of compound C is (100%) higher than that of compound A and B by relating to the ΔH_{vap} and intermolecular or intramolecular forces involved. (2)

3.2 Add column C to the table in such a way that ΔH_{vap} and ΔH_{fusion} are given in kJ mol⁻¹ and not in J mol⁻¹ for the diagram below.



- 3.2.1 Is water (100% of) or (100% of) the water (100%)? (2)
- 3.2.2 Explain the answer to QUESTION 3.1.1 about the volatility in the intermolecular forces involved. (2)

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Physical Science Paper Intermolecular Forces Investigation:

History of Science as a Facilitator for the Study of Physics Roberto Angeloni, 2019-01-10 This book serves to enhance scientific and technological literacy by promoting STEM Science Technology Engineering and Mathematics education with particular reference to contemporary physics The study is presented in the form of a repertoire and it gives the reader a glimpse of the conceptual structure and development of quantum theory along a rational line of thought whose understanding might be the key to introducing young generations of students to physics The recurrent theme here is that the conceptual extension of the concept of natural radiation symbolized by the constant h allows an easy method of charting the conceptual development of quantum theory The repertoire focuses on some momentous events of quantum theory including the discovery of the constant h which is one of the fundamental constants of nature and the key to understanding quantum mechanics the discovery of the photon by Albert Einstein and Niels Bohr's model of the hydrogen atom the experiments which led to disclosing the structure of atomic nuclei in the 1930s and the discovery of quantum mechanics and quantum electrodynamics which constitute the basis of contemporary particle physics Investigating Chemistry Matthew Johll, 2006-03-17 Matthew Johll's book introduces students from a non science background to the fundamentals of chemistry through an array of examples and applications from real life crime scenes Sherlock Holmes stories and authentic accounts of drug deals murders and thefts *Plenary and Invited Lectures* Milton Kerker, Albert C. Zettlemoyer, Robert L. Rowell, 2013-09-11 *Colloid and Interface Science Volume I Plenary and Invited Lectures* contains papers presented at the International Conference on Colloids and Surfaces held in San Juan Puerto Rico 21-25 June 1976 It consists of the plenary and invited papers and a general overview of these papers by A M Schwartz These papers were given during the morning sessions The volume is organized into 10 parts Part I contains papers on surface forces Parts II and III present studies on catalysis and aerosols respectively Part IV examines solid surfaces focusing on newer techniques for exploring surface structure and surface reactions The papers in Part V deal with water at interfaces including a lecture on the behavior and structure of water at inorganic surfaces including metals oxides and silicates Part VI covers the rheology of disperse systems including papers on the effect of inertial forces on the motion of solids through liquids and theoretical studies on diffusive heat flux Part VII takes up stability and instability in disperse systems steric stabilization and colloidal stability Parts VIII and IX examine biological membranes and surface thermodynamics respectively Part X on liquid crystals includes discussion of the structures and properties of this state of matter **Technical News Bulletin of the National Bureau of Standards** United States. National Bureau of Standards, 1968 **Technical News Bulletin**, 1970 *Dimensions*, 1951 **The Chemical News : and Journal of Physical Science**, 1894 **Progress in International Research on Thermodynamic and Transport Properties** Eric F. Lye, Joseph F. Masi, Roger Eichorn, 2013-10-22 *Progress in International Research on Thermodynamic and Transport Properties* covers the proceedings of the 1962 Second Symposium by the same title held at

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Mechanics Reviews ,1970 **The Molecular Vision of Life** Lily E. Kay,1993 This fascinating study examines the rise of American molecular biology to disciplinary dominance focusing on the period between 1930 and the elucidation of DNA structure in the mid 1950s Research undertaken during this period with its focus on genetic structure and function endowed scientists with then unprecedented power over life By viewing the new biology as both a scientific and cultural enterprise Lily E Kay shows that the growth of molecular biology was a result of systematic efforts by key scientists and their sponsors to direct the development of biological research toward a shared vision of science and society She analyzes the motivations and mechanisms empowering this vision by focusing on two key institutions Caltech and its sponsor the Rockefeller Foundation Her study explores a number of vital sometimes controversial topics among them the role of private power centers in shaping scientific agenda and the political dimensions of pure research It also advances a sobering argument the cognitive and social groundwork for genetic engineering and human genome projects was laid by the American architects of

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