

# Uniqueness of Carbon:

- · Basic building block of organic compounds
- · Permits long chains and strong carbon bonds to form (catenation)
- · Has different allotropes
- · Maximum 4 bonds w/ other elements
- · Graphite only non-metal that can conduct electricity

# Responsible to the second

#### Representation of Carbon

- 1. Molecular Formula: C2+1+O2
- 2. Empirical Formula: CH20 (simplest vatio)
- 3. Condensed / Semi-structural Formula: CH3 COOH of CH3-C
- 4. Structural Formula: H+C-C

# ENaming Rules

1. Find the longest chain and name the molecule from the no. of C atoms:

- 2. Single bonds or double bonds? } numbering : double bond has smallest #
- 3. Side chains : yl & lowest #
- 4. More than side chains of the same kind : 2 => di

commens, commas from words with - ,

3⇒tri 4⇒Ktra

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